

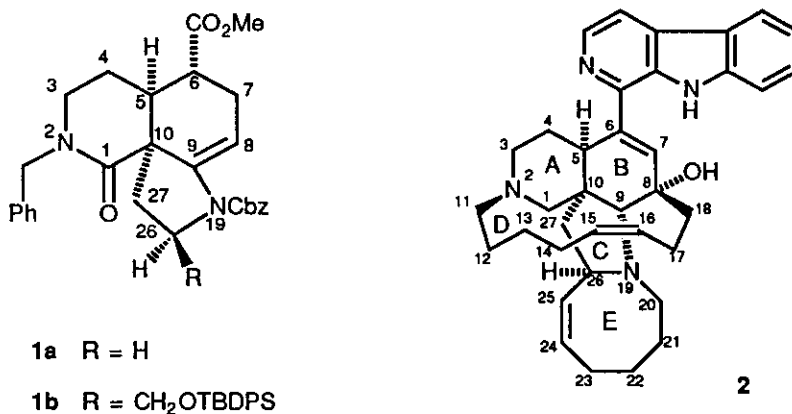
SYNTHESIS OF AN OPTICALLY ACTIVE TRICYCLIC INTERMEDIATE FOR MANZAMINES[#]

Karel M. J. Brands¹ and Upendra K. Pandit^{*}

Organic Chemistry Department, University of Amsterdam
Nieuwe Achtergracht 129, 1018 WS Amsterdam, The Netherlands

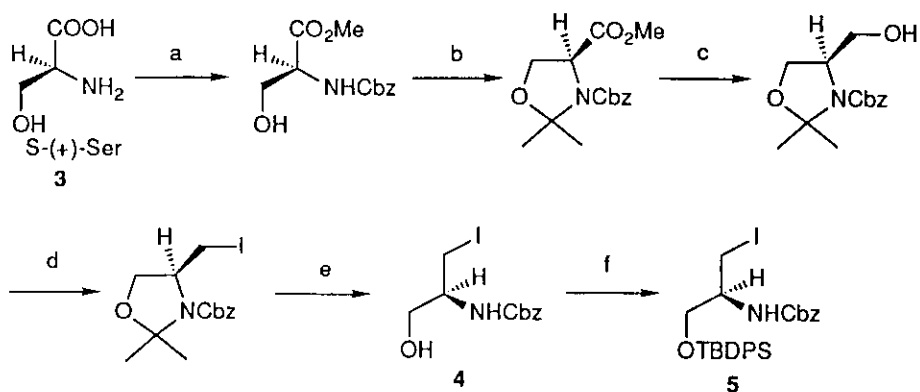
ABSTRACT - *L*-Serine has been converted into a chiral pyrrolo[2,3-*i*]isoquinoline derivative which can serve as a potential intermediate for manzamines.

In a recent communication we have presented a strategy for the synthesis of the tricyclic pyrrolo[2,3-*i*]isoquinoline system (**1a**)², which represents the ABC substructure of the manzamine alkaloids³ and, in addition, carries functional groups, which hold potential for the construction of the β -carboline and the thirteen-membered ring of our first synthetic target, namely, the alkaloid manzamine-A (**2**).



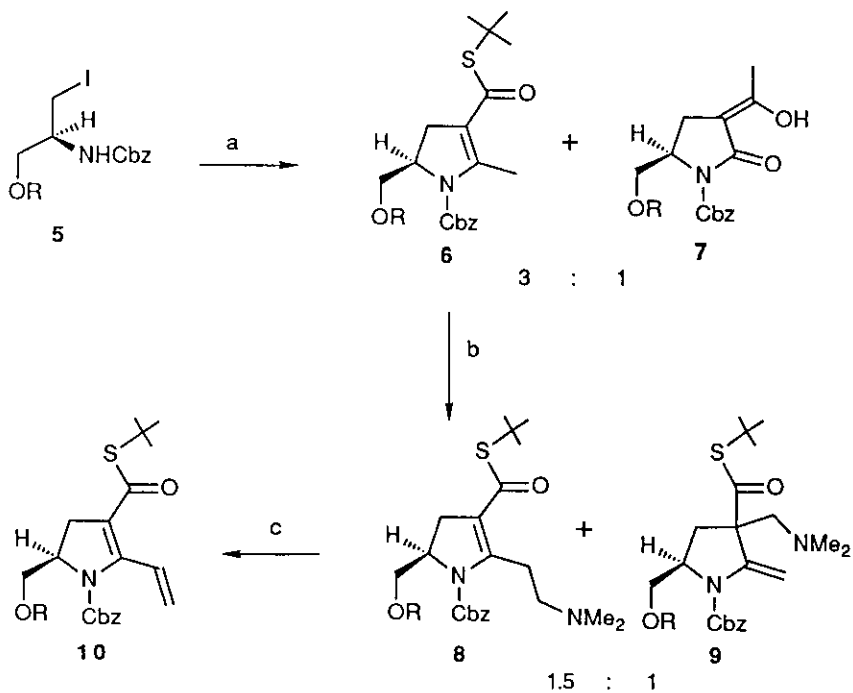
Another approach to the parent decahydropyrrolo[2,3-*i*]isoquinoline system has recently been reported by Hart.⁴ In this communication we describe the application of our strategy for the preparation of optically active tricyclic intermediate **1b**. It should be emphasized at the outset that the **5S, 10S, 26R**

[#] Dedicated to the memory of Professor Tetsuji Kametani; a dedicated and inspiring chemist and a good friend.



(a) i) SOCl_2 , MeOH; ii) CbzCl, NaHCO_3 ; 95%. (b) dimethoxypropane, TsOH; 99%.
 (c) $\text{Ca}(\text{BH}_4)_2$, EtOH/THF; 99%. (d) Ph_3P , I_2 , imidazole; 85%. (e) conc. HCl, acetone; 99%.
 (f) TBDPSCI, imidazole, DMF; 90%.

Scheme 1

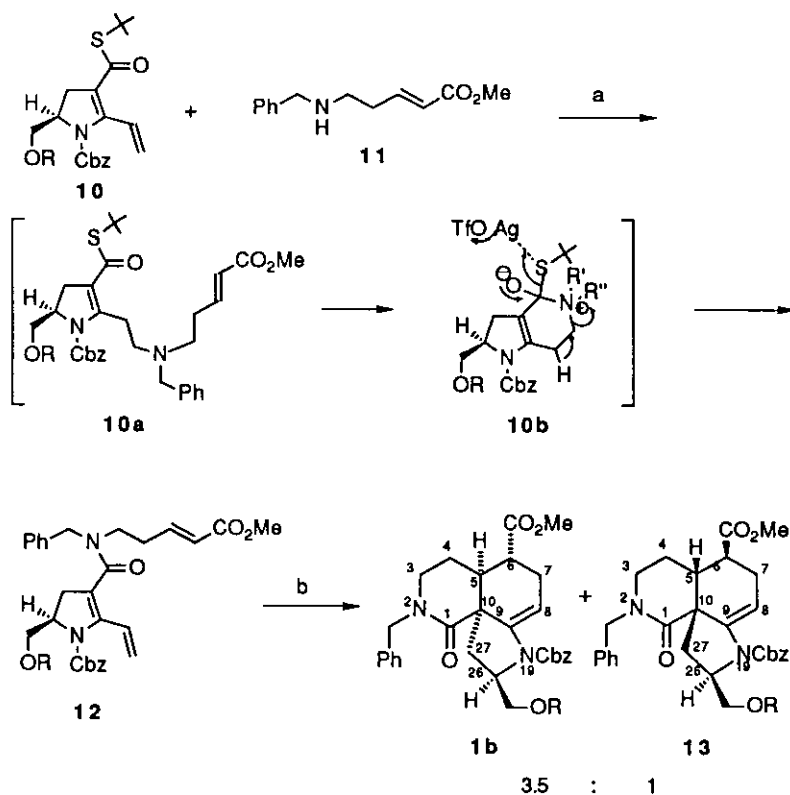


R = *tert*-butyldiphenylsilyl

(a) i) NaH, DME, *tert*-butylacetothioacetate; ii) TsOH, quinoline; 67%. (b) i) LiHMDS, THF, 6;
 ii) $\text{CH}_2=\text{N}(\text{CH}_3)_2^+ \Gamma^-$; 71%. (c) i) MeI, MeCN; ii) DBU, CH_2Cl_2 ; 71%.

Scheme 2

stereochemistry of **1b**⁵ corresponds to that of the natural alkaloid (**2**) and moreover, the hydroxymethyl group attached at C₂₆ is ideally suited for the construction of the azocine ring E. A retrosynthetic analysis of **1b**, along the lines described for **1a**, led to the requirement of optically pure iodide **5**. This compound was obtained in seven steps, starting from L-serine (**3**), in 71 % overall yield (**Scheme 1**).⁶ The optical purity of intermediate **4** was demonstrated with the aid of (+)-Eu(hfc)₃ shift reagent. After alkylation of *tert*-butylacetothioacetate with **5** and cyclization of the resulting product mixture, the desired pyrroline thiolester **6** could be isolated together with the γ -lactam **7** (**Scheme 2**). After much experimentation it was found that for a high yield condensation of Eschenmoser's salt with the anion of **6** the use of lithium hexamethyldisilazide in tetrahydrofuran was crucial. In addition to the desired product **8**, α -alkylation product **9** was also formed. Transformation of **8** to the rather unstable **10** proceeded straightforward. This compound was immediately coupled with aminoester **11**⁷ in the presence of silver triflate and diisopropylethylamine, yielding triene **12** in 69 % yield

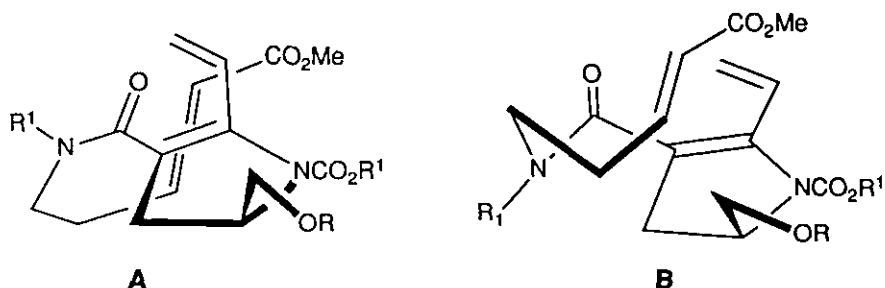


(a) AgOTf, DIPEA, MeCN; 69%. (b) xylene, Δ , 2h; 90%.

Scheme 3

(Scheme 3). This reaction proceeds via intermediates **10a** and **10b**. A Michael type addition product, corresponding to **10a**, has been isolated in high yield in the closely related coupling reaction of a model 2-pyrroline thiolester² and aminoester **11**. Subjection of this addition product to silver triflate yielded the expected amide coupling product. The fact that a corresponding aminolysis of thiolester **6**, lacking an activated vinyl group, under identical conditions did not lead to any reaction, further substantiates the proposed mechanism. Implications of this mechanism for the synthesis of **12** are being studied in our laboratory at the moment.

The intramolecular Diels-Alder reaction of **12** gave two diastereomeric products (3.5 : 1 ratio), to which structures **1b** and **13** have been assigned respectively, in 90 % combined yield. The diastereomeric transition states **A** and **B** can be envisaged for this reaction. As expected, the main and desired product (**1b**) is formed via the sterically more favourable transition state (**A**).



The gross structure elucidation of **1b** and **13** was facilitated by the corresponding data obtained for compound **1a**. The connectivities were clarified by COSY experiments. A clue for the distinction between **1b** and **13** is given by the chemical shifts of the protons attached to C₂₇. In compound **1b** H_{27exo} and H_{27endo} can be found at 1.61 and 2.64 ppm respectively, whereas in compound **13** these protons are observed at 1.97 and 2.23 ppm respectively. Steric repulsion between the silyloxymethylene group and the lactam carbonyl group in compound **1b** causes H_{27endo} to move into the deshielding cone of the lactam carbonyl group and H_{27exo} to move out of the shielding cone of the enecarbamate double bond, compared to the corresponding protons in compound **13**. In addition, the dihedral angle between H_{27endo} and H₂₆ in compound **1b** becomes 90°, which causes the former to be found as a doublet (J = 13.2 Hz). In compound **13** H_{27endo} is found as a doublet of a doublet (J = 12.4 and J = 7.2 Hz). The stereochemistry at C₂₅ in compound **1b** was also directly de-

duced from NOE experiments. Irradiation of H_5 at 1.95 ppm gave an enhancement for H_7 at 2.34 ppm and for $H_{27\text{exo}}$ at 1.61 ppm. Irradiation of $H_{27\text{endo}}$ at 2.64 ppm gave an enhancement for one of the protons of the silyloxymethylene group attached to C_{26} , found at 4.30 and 4.45 ppm. Conversely, irradiation of both methylene protons at 4.30 and 4.45 ppm gave only an enhancement for $H_{27\text{endo}}$ at 2.64 ppm.

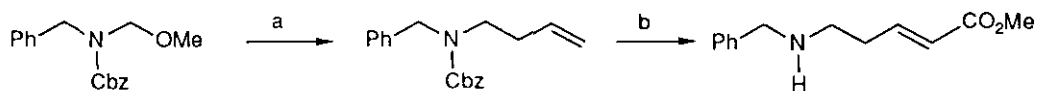
Compound **1b** is suitably functionalized at the centres N_2 , C_8 ; C_6 ; and C_{26} , N_{19} for the construction of the remaining rings of manzamine-A. Work towards this end is currently in progress.

ACKNOWLEDGEMENTS

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REFERENCES AND NOTES

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4. D. J. Hart and J. A. McKinney, *Tetrahedron Lett.*, **1989**, *30*, 2611.
5. Manzamine-A numbering according to reference 3d.
6. All new compounds reported herein have spectral (250-MHz ^1H nmr, ir and ms) data consistent with the assigned structures.
7. Synthesis of aminoester **11**, as reported in reference 2, has been slightly modified:



(a) allyltrimethylsilane, $\text{BF}_3 \cdot \text{Et}_2\text{O}$; quant. (b) i ozonolysis, ii Wittig reaction, iii deprotection; 60%.

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