

ALKALOIDS OF THALICTRUM. XIX.¹⁾
 REVOLUTOPINE AND THALIPINE, NEW APORPHINE-
 BENZYLISOQUINOLINE DIMERS FROM
THALICTRUM REVOLUTUM

Jinn Wu, Jack L. Beal, Wu-Nan Wu and Raymond W. Doskotch*

Division of Pharmacognosy and Natural Products Chemistry,
College of Pharmacy, Ohio State University
Columbus, Ohio 43210, U.S.A.

Revolutopine and thalipine, new alkaloids from the tops of Thalictrum revolutum have been assigned the aporphine-benzylisoquinoline structures 1 and 2, on the basis of physical and chemical evidence.

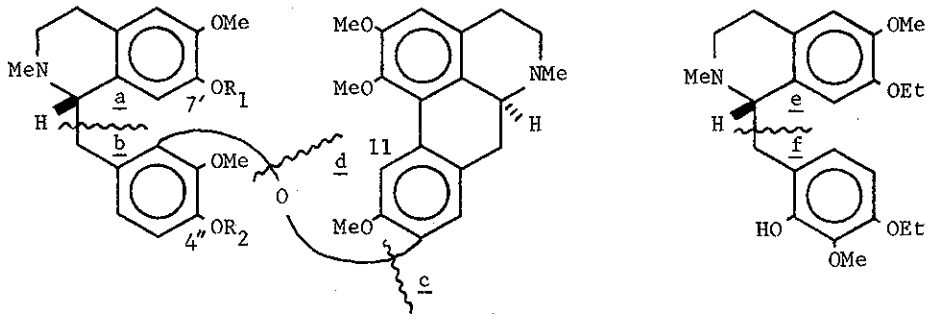
In an earlier study,²⁾ the roots of Thalictrum revolutum DC. had yielded thalicarpine (3). Recently, an analysis of the above-ground portion of this plant gave from the tertiary nonphenolic fraction seven alkaloids, and from the tertiary phenolic fraction eleven alkaloids of which two, revolutopine (1) and thalipine (2)³⁾ are new aporphine-benzylisoquinoline dimers present in very minor amounts.⁴⁾

Revolutopine (1) was isolated as an amorphous solid, $[\alpha]_D^{25} +126^\circ$ (c, 0.1 MeOH), $\lambda_{\max}^{\text{MeOH}}$ nm (log ϵ): 281 (4.40), 302 (4.24) and 314 shld

(4.15) with a bathochromic shift and hyperchromic effect under strong alkaline conditions to 284 (4.39) and 300 (4.39); and ir (KBr) band at ν_{\max} 3410 cm^{-1} (-OH) but no absorption in the carbonyl region. The nmr spectrum (60 MHz, CDCl_3 , TMS, δ in ppm) showed two N-methyl peaks at δ 2.29 and 2.43, five aromatic O-methyls at δ 3.65, 3.75, 3.83, 3.86 and 3.96, seven aromatic protons as five singlets at δ 6.33, 6.46 (2H), 6.56, 6.67 (2H), and 8.16 (characteristic of H-11 in aporphine),⁵⁾ and a broad two-proton singlet at δ 5.23 (D_2O exchanged) for the phenolic hydroxyls.

The spectral data pointed to revolutopine (1) as a diphenolic aporphine-benzylisoquinoline alkaloid with one phenolic group at C-7' since the characteristic C-7' O-methyl at $\delta \sim 3.58$ was lacking and the H-8' proton was present at δ 6.33, a position downfield from that ($\delta \sim 6.23$) expected for a proton ortho- to a methoxy.⁶⁾ In addition, the ms of revolutopine (1) showed a peak for fragment a ($R_1 = \text{H}$) at m/e 192 (100%), and for b ($R_2 = \text{H}$) at m/e 476. The molecular ion was observed at m/e 668 (1.5%) in agreement with the adopted formula, $\text{C}_{39}\text{H}_{44}\text{N}_2\text{O}_8$.

Treatment of revolutopine (1) with excess diazomethane produced the amorphous base 4 containing seven O-methyl groups, as the nmr spectrum (CDCl_3) exhibited peaks at δ 3.59, 3.67, 3.79, 3.81, 3.88 (double intensity) and 3.96. Comparison of the ir and nmr spectra, and the tlc mobility of 4 with O-methyl fetidine⁷⁾ showed them to be identical, thereby establishing the carbon skeleton and the oxygenation pattern for revolutopine (1). This alkaloid is only the second example of a fetidine-type aporphine-benzylisoquinoline dimer containing the 2,3,4-trioxygenated benzyl unit.⁸⁾ The configuration was determined as S,S on the basis that its cd spectrum (MeOH) showed maxima at $[\theta]_{305} -4,700$, $[\theta]_{275} -5,300$ and $[\theta]_{240} +43,000$ in

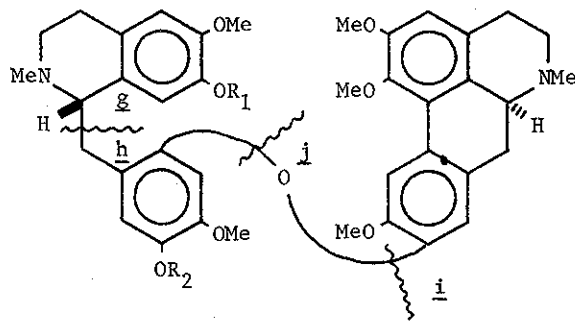


1, $R_1 = R_2 = H$

6

4, $R_1 = R_2 = Me$

5, $R_1 = R_2 = Et$



2, $R_1 = R_2 = H$

3, $R_1 = R_2 = Me$

7, $R_1 = H, R_2 = Me$

8, $R_1 = Me, R_2 = H$

agreement with those for (+)-fetidine (structure 4, with C-1 phenolic) at $[\alpha]_{305} -24,500$, $[\alpha]_{272} -14,200$ and $[\alpha]_{234} +148,000$.

The second phenolic group of revolutopine (1) was placed in the benzyl unit at C-4' since in the mass spectrum peaks at m/e 324 (9%, fragment c) and 340 (9%, fragment d) require that the aporphine unit contain three O-methyls, and a negative Gibbs' indophenol test⁹⁾ demands a para-substituted phenol. Confirmation of this assignment was obtained from the di-O-ethyl derivative 5, prepared from revolutopine (1) and diazoethane, which exhibited ms peaks at m/e 220 (100%) for fragment a ($R_1 = Et$), 504 (4) for b ($R_2 = Et$), 324 (1) for c, 340 (1) for d and M^+ at 724 (0.7). Sodium-ammonia cleavage of the diethyl derivative 5 produced the phenolic base 6, $[\alpha]_D^{25} -116^\circ$ (c, 0.2 MeOH), with nmr (90 MHz, $CDCl_3$) peaks for two ethyl groups; the methyl triplets were centered at δ 1.36 and 1.47 ($J = 7Hz$) and the ethylene quartet of double intensity was located at δ 4.08 ($J = 7Hz$). In addition, there appeared an N-methyl at δ 2.59, two O-methyls at δ 3.78 and 3.80, an AB quartet ($J = 8Hz$) centered at δ 6.20 and 6.54 for the two aromatic ortho-protons, and two other aromatic protons as singlets at δ 6.48 and 6.59. The ms of base 6 showed a peak at m/e 401 (0.8%) corresponding to the accepted formula $C_{23}H_{31}NO_5$, base peak at m/e 220 for fragment e, and fragment f at m/e 181 (1.3%). Base 6 gave a positive Gibbs' indophenol test.⁹⁾

Thalipine (2) was isolated as an amorphous solid, $[\alpha]_D^{25} +141^\circ$ (c, 0.19 MeOH), uv λ_{max}^{MeOH} nm (log ϵ): 282 (4.49), 303 shld (4.34) and 316 shld (4.19) with a hyperchromic change in strong base to 282 (4.51) and 303 (4.49); ir ($CHCl_3$) ν_{max} 3550 cm^{-1} for unassociated hydroxyl; nmr (90 MHz, $CDCl_3$) peaks for two N-methyls at δ 2.34 and 2.47, five O-methyls at δ 3.68, 3.79 (double intensity), 3.88 and 3.96, and seven aromatic protons as five

singlets at δ 6.44, 6.49 (2H), 6.59 (2H), 6.81 and 8.17; and ms peaks at m/e 668 (0.9%, M^+), 192 (100, g for $R_1 = H$), 476 (3, h for $R_2 = H$), 324 (10, i) and 340 (38, j). The physical data suggested a phenolic aporphine-benzylisoquinoline dimer and methylation of thalipine (2) with diazomethane afforded a heptamethoxy product identical with thalicarpine (3) when compared directly with an authentic sample. The cd spectrum of thalipine showed the same pattern as thalicarpine¹⁰⁾ with maxima at $[\theta]_{302} -14,800$, $[\theta]_{271} -19,000$ and $[\theta]_{235} +214,000$. Thus, thalipine is a diphenolic analogue of thalicarpine with both phenolic groups in the benzylisoquinoline part.

Partial methylation of thalipine (2) with diazomethane formed two monomethylated products that were separated by careful adsorption chromatography and were shown to be thalmelatine (7)¹¹⁾ and pennsylvanine (8)¹²⁾ by direct comparison of spectra (ir and nmr) and tlc mobility with those of authentic samples. Thalipine (2) is the third known diphenolic thalicarpine-type alkaloid; the other two are thalictrogamine^{6a)} and pennsylvanamine.¹²⁾

ACKNOWLEDGMENTS

We thank the U.S. Public Health Service for research grant No. HL-07502 in support of this work. We are grateful to Mr. C. Weisenberger of the Chemistry Department for several mass spectra, to Dr. E. H. Fairchild for the 90 MHz nmr spectra, and to Prof. M. P. Cava for a sample of (+)-fetidine.

REFERENCES AND FOOTNOTES

1. For Part XVIII, see J. Wu, J. L. Beal, R. W. Dосkotch and W.-N. Wu, Lloydia, in press.
2. T. Tomimatsu, E. Vorperian, J. L. Beal and M. P. Cava, J. Pharm. Sci., 1965, 54, 1389.
3. Prof. M. Shamma has informed us of the isolation of the same alkaloid from T. polygamum and we have mutually agreed on the name.
4. Details of the isolation method and the known alkaloids identified will be published elsewhere. Briefly, the alkaloid fractions were obtained from the ethanolic residue by partition steps, then subjected to adsorption column chromatography.
5. M. Shamma, 'The Isoquinoline Alkaloids,' Academic, New York, 1972, p. 328.
6. a) M. Shamma and J. L. Moniot, Tetrahedron Lett., 1973, 775.
b) M. Shamma, S. S. Salgar and J. L. Moniot, ibid., 1973, 1859.
7. Prepared by methylation of (+)-fetidine with diazomethane.
8. M. P. Cava and K. Wakisaka, Tetrahedron Lett., 1972, 2309.
9. F. E. King, T. J. King and L. C. Manning, J. Chem. Soc., 1957, 563.
10. R. W. Dосkotch, P. L. Schiff, Jr., and J. L. Beal, Tetrahedron Lett., 1968, 4999.
11. N. M. Mollov and H. B. Dutschewska, Tetrahedron Lett., 1964, 2219.
12. M. Shamma and J. L. Moniot, Tetrahedron Lett., 1974, 2291.

Received, 29th January, 1977