

SYNTHESIS OF HETEROCYCLIC COMPOUNDS XXII¹. PREPARATION OF 6-ARYL-2-THIOXOPYRIMIDINES

José L. García Navío, Antonio Lorente, and José L. Soto*

Departamentos de Química Orgánica de las Universidades de Alcalá de Henares and Complutense, Madrid-3, Spain

Abstract- The reactions of different ethyl benzylidenecyanoacetates (I) and benzylidenemalononitriles (V) with thiourea in an alcohol-alkoxide medium, provide a new and simple method for the synthesis of 6-aryl-5-cyano-4-oxo-2-thioxohexahydropyrimidines (III) and 4-amino-6-aryl-5-cyano-2-thioxotetrahydropyrimidines (VI), respectively. Similarly, the reaction of ethyl benzylidenemalonate leads to 5-ethoxycarbonyl-4-oxo-6-phenyl-2-thioxohexahydropyrimidine (XI).

Several methods of synthesis of thioxopyrimidines from α -substituted β -alkoxyacrylonitriles are known¹⁻¹⁰. All they involve the reaction with ureas or thioacetamides to obtain 2- or 4-thioxopyrimidines, according to the case. In a recent paper¹¹, the synthesis of 4-oxo-2-thioxopyrimidines through condensation of ethyl cyanoacetate, aldehydes and thiourea, in a potassium carbonate-ethanol basic medium, was reported.

In this paper, the reaction of different ethyl benzylidenecyanoacetates (I) with thiourea (II) in ethanol-sodium ethoxide is studied, by the result of which a method to synthesize 6-aryl-5-cyano-4-oxo-2-thioxohexahydropyrimidines (III) was found. In the same way, the reaction of different benzylidenemalononitriles (V) with thiourea in isopropanol-sodium isopropoxide leads to 4-amino-6-aryl-5-cyano-2-thioxo-1,2,3,6-tetrahydropyrimidines (VI).

The melting points and yields of the compounds III and VI are summarized in Table 1.

TABLE 1

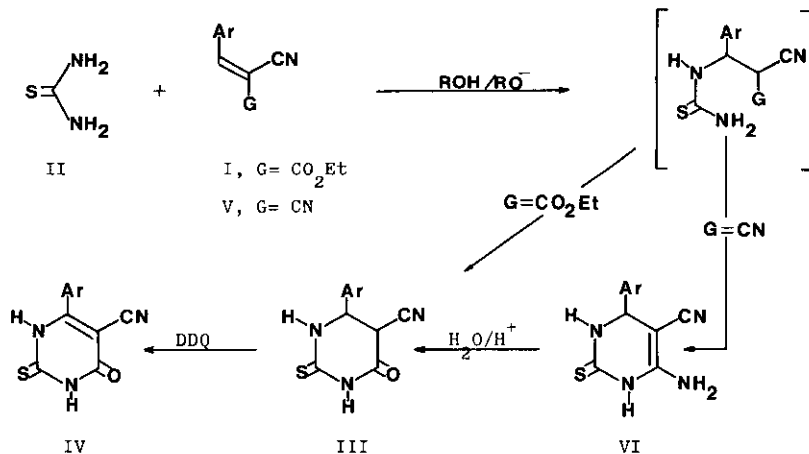
	Ar	mp (°C)	III	Yield (%)	mp (°C)	VI	Yield (%)
a	C ₆ H ₅	239-240		65	216-217		57
b	p-CH ₃ -C ₆ H ₄	244-245		82	215-216		63
c	p-CH ₃ O-C ₆ H ₄	231-232		50	204-206		35
d	m-NO ₂ -C ₆ H ₄	220-221		30	215-216		21

All products gave satisfactory microanalyses (C_± 0.30, H_± 0.34, N_± 0.29, S_± 0.21)

The formation of the pyrimidines III and VI takes place easily at room temperature and can be explained through condensation of thiourea with the alkylidene compound I or V, followed by cyclization of the adduct (Scheme 1) by attack of the amidic nitrogen to the ethoxycarbonyl or the cyano group, respectively. It should be pointed out that in the reaction between I and thiourea the formation of products through cyclization of the amidic nitrogen with the cyano group has not been observed.

The structure of the compounds obtained was established on the basis of spectral data (Tables 2 and 3) and, in some cases, by chemical tests. Thus, the oxidation of IIIc with dichlorodicyanoquinone (DDQ) yields the corresponding 4-oxo-2-thioxo-1,2,3,4-tetrahydropyrimidine IVc that was identified by comparison with the one obtained by Kambe et al.¹¹ The structure of the pyrimidines VI was confirmed by hydrolysis of VIa with dilute hydrochloric acid to 4-oxo-2-thioxohexahydropyrimidine IIIa.

Scheme 1

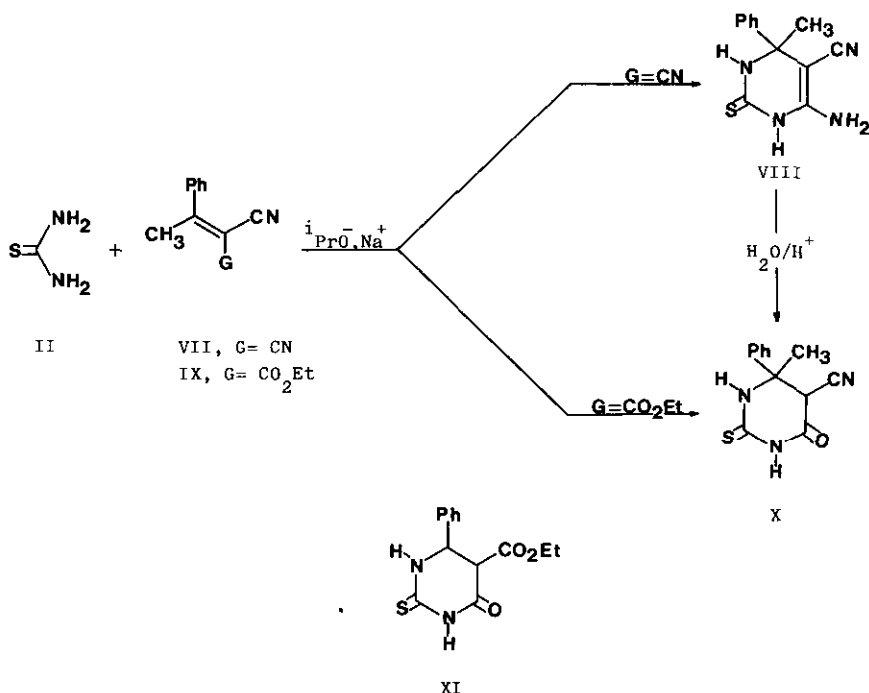


In the reaction of benzylidenemalononitriles V with thiourea, in an ethanol-sodium ethoxide medium, besides the 4-amino-2-thioxo-1,2,3,6-tetrahydropyrimidines VI, the corresponding 2-amino-4-aryl-3,5-dicyano-6-ethoxypyridines are also formed as side products. Their structure was identical to the ones obtained from benzylidenemalononitriles and malononitrile in an ethanol-sodium ethoxide system¹². The formation of those 6-ethoxypyridines can be explained through condensation of V and malononitrile originated from benzylidenemalononitrile in the basic reaction medium.

The reactions of ethyl β -methylbenzylidenecyanoacetate (IX) and β -methylbenzylidenemalononitrile (VII) with thiourea, in isopropanol-sodium isopropoxide, have also been investigated (Scheme 2). Thus, 5-cyano-6-methyl-6-phenyl-4-oxo-2-thioxohexahydropyrimidine (X) and 4-amino-5-cyano-6-methyl-6-phenyl-1,2,3,6-tetrahydropyrimidine (VIII) were obtained, respectively. Hydrolysis of VIII with concentrated hydrochloric acid in ethanol under reflux affords 4-oxo-2-thioxohexahydropyrimidine X with an almost quantitative yield.

Finally, the reaction of ethyl benzylidenemalonate with thiourea in isopropanol-sodium isopropoxide leads to 5-ethoxycarbonyl-4-oxo-6-phenyl-2-thioxohexahydropyrimidine (XI) in high yield.

Scheme 2



EXPERIMENTAL

All melting points were determined in open capillary on a Büchi SMP-20 and are uncorrected. IR spectra were performed on a Perkin-Elmer 700. Reported values are the more intense or characteristic peaks. ¹H-nmr spectra were recorded on a Varian T-60 A with TMS as internal standard. Mass spectra were registered on a Varian MAT 711. Column chromatographies were carried out on basic alumina 60 Merck.

Benzylidenemalononitriles and ethyl benzylidenecyanoacetates were prepared according to general methods described by Corson¹³ and Bertini¹⁴, respectively. Ethyl β-methylbenzylidenecyanoacetate was prepared according to the synthesis described by Cope¹⁵ and β-methylbenzylidenemalononitrile by the method of Mowry¹⁶. Ethyl benzylidenemalonate was obtained by the procedure of Kroeker and McElvain¹⁷.

6-Aryl-5-cyano-4-oxo-2-thioxohexahydropyrimidines III (General procedure).— To a solution of sodium (230 mg) in ethanol (50 ml), thiourea (760 mg, 10 mM) was added. When it became dissolved, 10 mM of the corresponding ethyl benzylidenecyanoacetate (I), dissolved in ethanol, were added. The resulting mixture was stirred at room temperature for 24 hr and after a solid precipitated through addition of 5% acetic acid. This solid was filtered and recrystallised from ethanol to obtain the corresponding hexahydropyrimidine III.

4-Amino-6-aryl-5-cyano-2-thioxo-1,2,3,6-tetrahydropyrimidine VI (General procedure).— To a solution of sodium (230 mg) in isopropanol (40 ml), thiourea (760 mg, 10 mM) and the corresponding benzyli-

denemalononitrile V (10 mM), dissolved in isopropanol, were successively added. After 36 hr (Note 1) the reaction mixture was precipitated with 5% acetic acid. The solid was filtered and washed with water. In every cases the different tetrahydropyrimidines VI were recrystallised from ethanol.

Experimental results and spectroscopic data of pyrimidines III and VI are summarized in Tables 1-3.

Hydrolysis of VIa: Preparation of 5-cyano-4-oxo-6-phenyl-2-thioxohexahydropyrimidine (IIIa).-230 mg (1 mM) of VIa were suspended in ethanol (10 ml) and the solution was acidified with 10 % hydrochloric acid (10 ml). The mixture was kept with stirring at room temperature for 12 hr and later the precipitate was filtered, washed with water, dried and recrystallised in ethanol, giving 220 mg of IIIa. Yield: 96%; mp 238-239 °C.

Oxidation of IIIc: Preparation of 5-cyano-6-(p-methoxyphenyl)-4-oxo-2-thioxo-1,2,3,4-tetrahydropyrimidine (IVc).- 522 mg (2 mM) of IIIc were suspended in toluene (50 ml) and later DDQ (680 mg, 3 mM) was added. The mixture was refluxed for 8 hr and the resulting solid was filtered and chromatographed in a basic alumina column. IVc was obtained by concentrating the fractions that contained it and further recrystallisation from ethanol. Yield: 43%; mp 281-283 °C. IR (KBr): $\nu_{\max} = 3200, 3100, 2940, 2230, 2220, 1670, 1600, 1560, 1510, 1460, 1310, 1265, 1220, 1180 \text{ cm}^{-1}$. $^1\text{H-NMR}$ (DMSO- d_6): $\delta = 12.82-12.52$ (b. s., 2 H, 1 and 3-H); 7.46, 7.33, 6.93, and 6.80 (q, 4H arom., syst. AB, $J = 7.8$ Hz); 3.73 (s, 3H, OCH_3). MS: m/e (relative intensity) = 259 (M^+ , 100); 258 (31); 231 (16); 201 (18); 158 (15); 134 (32); 77 (4).

4-Amino-5-cyano-6-methyl-6-phenyl-2-thioxo-1,2,3,6-tetrahydropyrimidine (VIII).- To a solution of sodium (345 mg) in isopropanol (40 ml), thiourea (1.14 g, 15 mM) and β -methylbenzylidenemalononitrile (2.52 g, 15 mM) were added in successive fractions. The mixture was stirred at room temperature for 36 hr. Ether (15 ml) was added before acidification with 5% acetic acid up to a slightly acidic pH. The precipitate was filtered and washed several times with water and ether. The crystallisation in ethanol with charcoal afforded 1.3 g of VIII. Yield: 36%; mp 249 °C. IR (KBr): $\nu_{\max} = 3400, 3300, 3210, 3150, 3000, 2195, 1660, 1575, 1470, 1445, 1380, 1265, 1175, \text{cm}^{-1}$. $^1\text{H-NMR}$ (DMSO- d_6): $\delta = 9.73$ (s, 1H, N-H); 9.60 (s, 1H, N-H); 7.13 (s, 5H arom.); 5.90 (b. s., 2H, NH_2); 1.65 (s, 3H, CH_3). Anal. Calcd. for $\text{C}_{12}\text{H}_{12}\text{N}_4\text{S}$: C, 58.99; H, 4.95; N, 22.93; S, 13.12. Found: C, 58.82; H, 5.06; N, 22.89; S, 13.29 %.

5-Cyano-6-methyl-4-oxo-6-phenyl-2-thioxohexahydropyrimidine (X) (Procedure A).- To a solution of sodium (230 mg) in isopropanol (30 ml), thiourea (760 mg, 10 mM) was added, followed by slow addition of ethyl β -methylbenzylidenecyanoacetate dissolved in dry isopropanol (30 ml). The mixture was stirred at room temperature for 48 hr. Later it was diluted with ether (15 ml) and acidified with 5% acetic acid. The resulting solid was washed with water and ether, dried and crystallised from ethanol obtaining 1.5 g of X. Yield: 61%; mp 216-217 °C. IR (KBr): $\nu_{\max} = 3125, 3050, 2930, 2250, 1730, 1550, 1490, 1440, 1380, 1335, 1235, 1175, 1150, 1120 \text{ cm}^{-1}$. $^1\text{H-NMR}$ (DMSO- d_6): $\delta = 11.84$ and 11.76 (d, 1H, N-H); 10.60 (s, 1H, N-H); 7.37 (s, 5H arom.); 5.34 and 5.05 (1H, 5-H); 1.81 and 1.68 (3H, CH_3). Anal. Calcd. for $\text{C}_{12}\text{H}_{11}\text{N}_3\text{OS}$: C, 58.75; H, 4.52; N, 17.13; S, 13.07. Found: C, 59.09; H, 4.72; N, 17.31, S, 13.36.

Hydrolysis of 4-amino-5-cyano-6-methyl-6-phenyl-2-thioxo-1,2,3,6-tetrahydropyrimidine (VIII) (Procedure B).- 415 mg (17 mM) of VIII in ethanol (40 ml) were acidified with concentrated hydrochloric

Table 2. IR (KBr, cm^{-1})

	III	VI
a	3250; 3130; 2250; 1715; 1540; 1235.	3370; 3300; 3170; 2180; 1650; 1580; 1475.
b	3125; 2920; 2250; 1700; 1560; 1230.	3380; 3300; 3170; 2180; 1650; 1580; 1475.
c	3100; 2940; 2240; 1700; 1550; 1235.	3300; 3170; 2180; 1650; 1580; 1500; 1470.
d	3300; 3120; 2240; 1700; 1545; 1220.	3290; 3180; 2185; 1650; 1570; 1540; 1480.

Table 3. $^1\text{H-NMR}$ (DMSO-d_6 , ppm)

	$\text{N}^3\text{-H}$	$\text{N}^1\text{-H}$	H-C^5 and C^6	Aromatic	
IIIa	11.66-11.36 (b.s.)	10.52-10.32 10.06-9.86	5.30-4.73 (m)	7.27 (s)	
IIIb	11.60-11.33 (b.s.)	10.23-10.03 9.97-9.77	5.20-4.68 (m)	7.23-6.93 (m)	
IIIc	11.68-11.38 (b.s.)	10.32-10.12 10.02-9.82	5.22-4.72 (m)	7.45-6.75 (m)	
IIId	11.79-11.46 (b.s.)	10.39-10.16 10.16-9.96	5.56-5.42 (m)	8.60-7.60 (m)	
	$\text{N}^3\text{-H}$	$\text{N}^1\text{-H}$	$\text{C}^6\text{-H}$	Amino	Aromatic
VIa	9.68 (s)	9.41+9.36 (d, J=3 Hz)	4.93+4.88 (d, J=3 Hz)	6.00 (s)	7.18 (s)
VIb	9.70 (s)	9.45+9.40 (d, J=3 Hz)	4.87+4.82 (d, J=3 Hz)	5.98 (s)	6.98 (s)
VIc	9.65 (s)	9.38+9.33 (d, J=3 Hz)	4.82+4.77 (d, J=3 Hz)	5.92 (s)	7.03+6.90+ 6.78+6.65 (q)
VIId	9.90 (s)	9.65+9.60 (d, J=3 Hz)	5.21+5.16 (d, J=3 Hz)	6.13 (s)	8.20-7.33 (m)

acid (6 ml). The mixture was heated under reflux for 10 hr and afterwards the ethanol was removed up to half volume. Through addition of water a solid precipitated, which was filtered, washed with water and recrystallised from ethanol, giving 417 mg of X. Yield: 98%.

5-Ethoxycarbonyl-4-oxo-6-phenyl-2-thioxohexahydropyrimidine (XI).— To a solution of sodium (230 mg) in dry isopropanol (35 ml), thiourea (760 mg, 10 mM) and ethyl benzylidenemalonate (2.48 g, 10 mM) were added. The mixture was stirred at room temperature for 24 hr and the solvent was then removed. To the resulting residue ethanol (5 ml) and water, in an amount just to obtain a clear solution, were added. This solution was acidified with 5% acetic acid up to a slightly acidic pH. The white precipitate appeared was kept in freezer for 1 hr. The solid obtained was washed several times with water and recrystallised from ethanol, giving 2.16 g of a white product, whose several recrystallisations fixed its melting point at 170 °C. Yield: 78%. IR (KBr): ν_{max} = 3180, 2990, 1735, 1715, 1565, 1440, 1350, 1300, 1270, 1210, 1170, 1110, 1025 cm^{-1} . $^1\text{H-NMR}$ (DMSO-d_6): δ = 11.15 (b. s., 1H, 3-N); 9.87 (b. s., 1H, 1-N); 7.15 (s, 5H arom.); 5.05–4.73 (m, 1H, C-H); 4.2, 4.06, 3.97, and 3.85 (q, 2H, CH_2); 1.2, 1.08, and 0.98 (t, 3H, CH_3); 1.01 (1H, C-H).

Anal. Calcd. for $\text{C}_{13}\text{H}_{14}\text{N}_2\text{O}_2\text{S}$: C, 56.10; H, 5.07; N, 10.07; S, 11.52. Found: C, 56.42; H, 5.30; N, 10.03; S, 11.66.

ACKNOWLEDGEMENTS

We are indebted to Drs. García Martínez and Seoane Prado for measurements of MS and $^1\text{H-NMR}$ spectra and to Cristobal López Pérez for his collaboration.

REFERENCES AND NOTES

- 1 Previous paper: J. L. Soto, A. Lorente and J. L. García Navío, An. Quim., **77** C, 255 (1981)
- 2 H. Sutter and E. Habicht, U. S. Pat. 2,698,326 (28 Dec., 1954); Chem. Abstr., **50**, 1093 e (1956).
- 3 E. C. Taylor, R. J. Knopf, J. A. Cogliano, J. W. Barton and W. Pfleiderer, J. Am. Chem. Soc., **82**, 5711 (1960).
- 4 S. G. Cottis and H. Tieckelmann, J. Org. Chem., **26**, 79 (1961).
- 5 R. Granados and C. V. Sala, An. Real Soc. Esp. Fis. Quim., **42**, 349 (1946).
- 6 W. J. Middleton and V. A. Engelhardt, J. Am. Chem. Soc., **80**, 2829 (1958).
- 7 E. C. Taylor and McKillop, "The Chemistry of Cyclic Enaminonitriles and o-Aminonitriles", Wiley (Interscience), New York, 1970, p. 113.
- 8 T. S. Griffin, T. S. Woods and D. L. Klayman, Adv. Heterocyclic Chem., **18**, 127 (1975).
- 9 T. L. V. Ulbricht and C. C. Price, J. Org. Chem., **21**, 567 (1956).
- 10 A. Lorente and J. L. Soto, An. Quim., **76** C, 242 (1980).
- 11 S. Kambe, K. Saito, H. Kishi, A. Sakurai and H. Midorikawa, Synthesis, 287 (1979).
- 12 M. A. Cabrerizo and J. L. Soto, An. Quim., **70**, 951 (1974).
- 13 B. B. Corson and R. W. Stoughton, J. Am. Chem. Soc., **50**, 2825 (1928).
- 14 Bertini, Gazz. Chim. Ital., **31**, 279 (1901).
- 15 A. C. Cope, C. M. Hofmann, C. Wickoff and E. Hardenberg, J. Am. Chem. Soc., **63**, 3452 (1941).
- 16 D. T. Mowry, J. Am. Chem. Soc., **67**, 1050 (1945).
- 17 R. F. B. Cox, E. H. Kroeker and S. M. McElvain, J. Am. Chem. Soc., **56**, 1173 (1934).

Note 1. The formation of VI_d required 72 hr of reaction and the precipitate obtained, according to the general procedure, was purified by column chromatography with silicagel, followed by recrystallisation from ethanol.

Received, 2nd October, 1981