

SATURATED HETEROCYCLES, 47.¹ SYNTHESIS OF SOME TETRAHYDROISOQUINOLINE
CONDENSED 1,3-HETEROCYCLES

Ferenc Fülöp, M. Sami El-Gharib, Attila Sohajda, Gábor Bernáth^{*}

Institute of Pharmaceutical Chemistry, University Medical School
Szeged, Eötvös u. 6., Hungary

Jenő Kóbor

Chemical Department, Pedagogical High School Szeged, Árpilis 4. u. 6.,
Hungary

György Dombi

Institute of Organic Chemistry, József Attila University Szeged,
Dóm tér 8., Hungary

Abstract - Calycotomine (2) and homocalycotomine (3) reacted with isothiocyanates to give the corresponding thiocarbamides (4, 5). The latter were treated with methyl iodide, and subsequent elimination of methyl mercaptan on treatment with alkali afforded 1,3-oxazolo[4,3-*a*]-isoquinolines (6) and 1,3-oxazino[4,3-*a*]isoquinolines (7). 1,3-Thiazolo[4,3-*a*]isoquinolines (8) and 1,3-thiazino[4,3-*a*]isoquinolines (9) were prepared from 4 and 5 with hydrogen chloride. Compound 7a was also synthesized by the reaction of 3 with phenyl isocyanate, followed by the elimination of water; 9a was obtainable from 7a by treatment with P₄S₁₀. The structures of compounds 6-9 were confirmed by ¹H NMR.

In recent decades several guanidine derivatives have found medical use (e.g. Debrisoquine, Guanethidine, Clonidine).² Substitution of one of the guanidine nitrogens by a bioisosteric atom (O, S) gives compounds of high and varied pharmacological activities. Of the heterocyclic derivatives in this group, the main objects of research were the 2-imino-substituted 1,3-thiazoles, 1,3-thiazines, 1,3-oxazoles and 1,3-oxazines.³⁻⁷

Compounds 4 and 5 were allowed to stand for 1 h at room temperature with excess methyl iodide to give the thiuronium salts. In one case, when starting from 5a, the thiuronium salt was isolated (mp 154-156°C, from EtOH¹³). The residue obtained on evaporation of the methyl iodide-containing reaction mixture was stirred at room temperature in methanol containing 3 N potassium hydroxide. After methyl mercaptan had been completely expelled (2-4 h), the reaction mixture was evaporated. When R = C₆H₅, the residue was mixed with water and the crystalline product was filtered off; in the case of R = C₆H₁₁, the product was isolated by extraction of the evaporation residue with hot benzene. Both methods of work-up gave the 1,3-oxazolo[4,3-a]isoquinolines (6) and 1,3-oxazino[4,3-a]-isoquinolines (7), respectively, in good yields.

Refluxing of 4 and 5 for 15 min in ethanol containing 20% dry hydrogen chloride gave, after evaporation and neutralization, 1,3-thiazolo[4,3-a]isoquinolines (8) and 1,3-thiazino[4,3-a]isoquinolines (9), respectively.

The mp, IR and ¹H NMR data on the synthesized compounds are shown in the Table.

The 1,3-oxazine 7a was also prepared from homocalycotomine (3) and phenyl isocyanate *via* the urea derivative 10 (mp 167-168°C, from EtOH¹²), by treatment with thionyl chloride; however, the yield was only 17%.

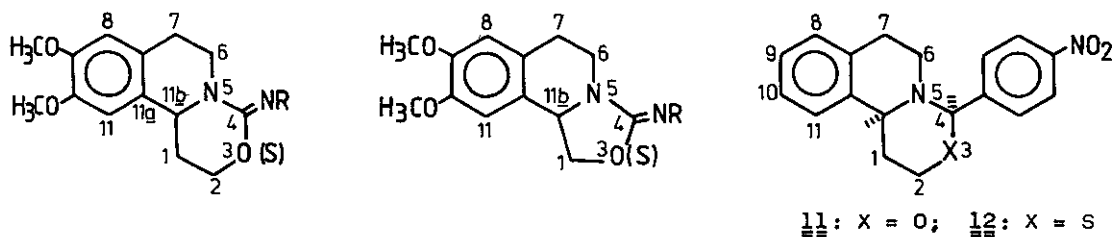
The 1,3-thiazine (9a) was synthesized in an alternative route¹⁴ by heating 7a with phosphorus pentasulphide for 2 h at 150°C; the product (9a) was obtained in 31% yield.

Crabb and Mitchell¹⁵ synthesized and made a detailed ¹H NMR study of the closely analogous compounds, *cis*-4-(*p*-nitrophenyl)-1,6,7,11b-tetrahydro-2H,4H-1,3-oxazino[4,3-a]isoquinoline (11) and *cis*-4-(*p*-nitrophenyl)-1,6,7,11b-tetrahydro-2H,4H-1,3-thiazino[4,3-a]isoquinoline (12). In the oxazino 11 and in the related thiazino derivative 12 Bohlmann bands were absent and the NMR spectrum indicated predominance of the O (or S) inside form as concerns the two possible B/C *cis* conformers.

The ¹H NMR spectra of the compounds 6-9 synthesized in the present work are in good agreement with the data measured by Crabb (*cf.* Table). As the lone electron pair of 5-N is considerably conjugated with the near-by hetero atoms, the appearance of Bohlmann bands cannot be expected. The presence of $\begin{array}{c} \text{N}-\text{C}=\text{N} \\ | \\ \text{O} \end{array}$ or $\begin{array}{c} \text{N}-\text{C}=\text{N} \\ | \\ \text{S} \end{array}$ groups makes the conformation of the hetero ring near-planar. This is

No	Mp (°C) ^b	IR ¹⁶ ν _{max} (cm ⁻¹)	1H NMR ¹⁷ data ^a				Coupling constants (Hz)			
			Chemical shifts (ppm)							
			OCH ₃	-CH ₂ -O(s)	H-11b	H-11	H-8	J _{11b,1ax}	J _{11b,1eq}	
<u>6a</u>	129-130	1680	3,82	4,6	4,15	6,47	6,65	7,5	4,0	
<u>6b</u>	104-105	2910, 1675	3,83	4,7	4,15	6,55	6,65	7,5	4,0	
<u>7a</u>	115-118	1585, 1550	3,82	4,6	4,15	6,6	6,66	8,0	4,0	
<u>7b</u>	128-129	2910, 1620	3,84	4,5	4,15	6,61	6,64	8,0	4,5	
<u>11c</u>	145-147			4,03	4,17			11,9	4,0	
<u>8a</u>	147-148	1625	3,83	4,7	3,5	6,6	6,68	10,0	6,0	
<u>8b</u>	134-135	2905, 1605	3,82	4,7	3,55	6,62	6,65	10,0	6,0	
<u>9a</u>	158-160	1575	3,84	4,7	4,40	6,63	6,67	9,0	4,5	
<u>9b</u>	111-112	2915, 1575	3,84	4,6	4,35	6,61	6,66	9,0	5,0	
<u>12c</u>	114-116			4,15	4,24			11,5	3,5	

^a In order to facilitate comparison, the numbering of compounds 6-12 in the Table is as follows:



^b All compounds were recrystallized from ethanol.

^c Compounds synthesized and studied by Crabb and Mitchell.¹⁵

supported by the higher chemical shifts of the $\text{CH}_2\text{-O(S)}$ protons as compared with the corresponding protons in compounds 11 or 12, and also by the $J_{11b,1ax}$ values, which are significantly smaller than in usual diaxial couplings.

Results of the current investigation of some reactions of 6-9 and a high-resolution NMR study of these compounds will be described in a forthcoming publication.

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16. IR spectra were recorded with a SPECORD 75 IR instrument in KBr pills.
17. ^1H NMR spectra were recorded at 60 MHz with a JEOL C60 spectrometer at room temperature in CDCl_3 solution with TMS as internal standard.

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