

HETEROCYCLES, Vol. 89, No. 9, 2014, pp. 2105 - 2121. © 2014 The Japan Institute of Heterocyclic Chemistry
Received, 30th July, 2014, Accepted, 26th August, 2014, Published online, 28th August, 2014
DOI: 10.3987/COM-14-13065

HYPERVALENT IODINE MEDIATED ONE-POT C-H FUNCTIONALIZATION AT 2 α - OR 3 α -POSITION OF INDOLE DERIVATIVES

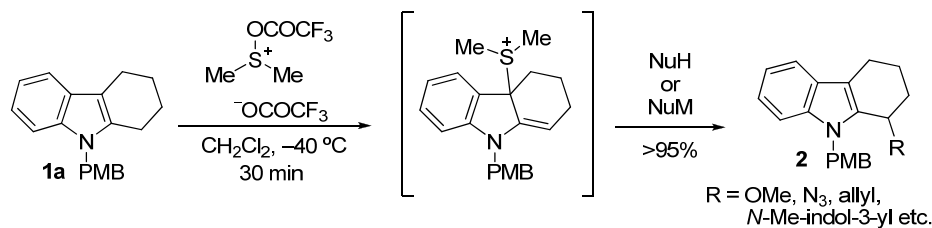
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Abstract – The one-pot 2 α - and 3 α -functionalization of 2,3-disubstituted indoles using a hypervalent iodine reagent has been developed. The substitution at the 2 α -position of indoles took place using phenyliodinebis(trifluoroacetate) with oxygen and carbon nucleophiles in moderate yields. The combination of iodosobenzene and trimethylsilyl azide afforded 3 α -azide derivatives preferentially. The latter reaction was applied to other 2,3-disubstituted indoles.

INTRODUCTION

Hypervalent iodine reagents¹ having a strong leaving-group ability and electrophilicity allow a wide range of reactions, such as oxidations and C-C coupling reactions under mild conditions with a tolerance for a wide range of other functional groups. The reaction between arenes and hypervalent iodine is attractive for one of the interesting research area.² Especially the application of the electron-rich arenes such as indoles give rise to the substitution via cation radical intermediate,³ indolyliodonium fragmentation,⁴ and so on.⁵ Recently, Ishibashi's group revealed the combination of phenyliodine diacetate (PIDA) and tetrabutylammonium iodide (TBAI) reacted with tetrahydrocarbazoles to give 2 α -acetoxy derivatives.⁶ Alternatively, Du Bois and co-workers developed that the reaction of tetrahydrocarbazoles using imino λ^3 -iodanes with a rhodium catalyst affords 2 α - and 3 α -amino derivatives.⁷ Recently, we developed the concise 2 α -functionalization of indole derivatives using a thionium species generated from DMSO-TFAA (Scheme 1).⁸



Scheme 1. Thionium mediated 2 α -functionalization of indole derivatives

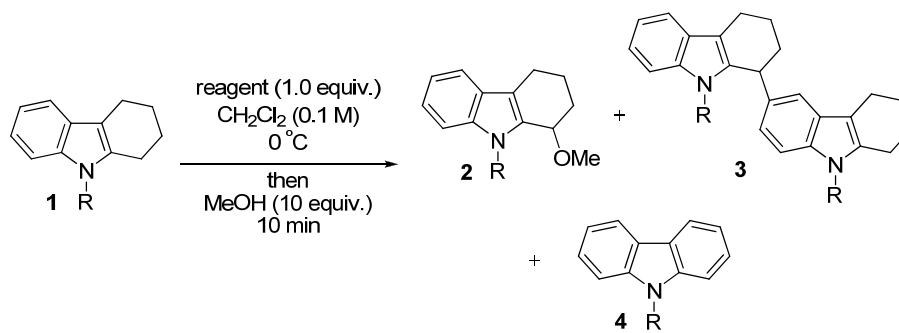
This method is useful for the introduction of various substituents to the 2 α -position⁹ in one-pot procedure under mild reaction conditions. However, we require a more practical method to access to 2 α -functionalized indoles for the synthesis of biologically active compounds.¹⁰ We report herein the 2 α - and 3 α -functionalization of indole derivatives using a hypervalent iodine reagent.

RESULTS AND DISCUSSION

Optimization of Reaction Conditions

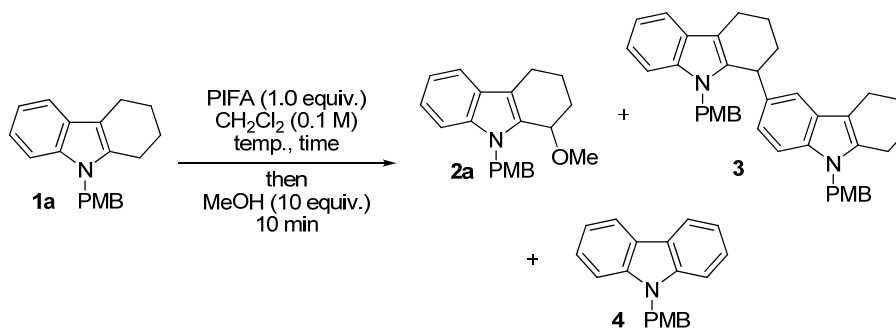
To obtain the desired 2 α -methoxy indole derivatives more conveniently, we investigated the reactivity of hypervalent iodine reagents toward *N*-*p*-methoxybenzyl (PMB)-tetrahydrocarbazole **1a** (Table 1). To a solution of **1a** in CH₂Cl₂ added one equivalent of hypervalent iodine reagent at 0 °C. After consumption of **1a**, ten equivalents of MeOH as a nucleophile was added to the reaction mixture. When Dess-Martin periodinane (DMP, entry 1) and phenyliodine diacetate (PIDA, entry 2) were used, carbazole **4** and dimer **3** was obtained respectively instead of the desired 2 α -methoxy product **2a**. PhI(OH)OTs (Koser's reagent)¹¹ afforded a trace amount of **2a** along with dimer **3** (22%) and carbazole **4** (5%), respectively (entry 3). The use of iodopentafluorobenzene bis(trifluoroacetate) (FPIFA) as a more electrophilic λ^3 -iodan produced **2a** and **3** in 8% and 29% yields, respectively (entry 4). In the case of phenyliodine bis(trifluoroacetate) (PIFA) gave **2a** in 35% yield (entry 5). The substituent effect at the indole nitrogen was examined for PIFA. *N*-Unsubstituted tetrahydrocarbazole gave a complex-mixture (entry 6) and Boc and acetyl derivatives produced a trace amount of 2 α -methoxy compounds **2c** and **2d** (entries 7, 8).

Subsequently, we examined the effect of reaction temperature using PIFA (Table 2). At room temperature, the reaction afforded no desired product **2a** and gave dimer **3** in 43% yield (entry 1). When the reaction was performed at -20 °C, **2a** was given in 30% yield (entry 3). At -30 °C, the 2 α -methoxy compound **2a** was obtained in 24% yield without the formation of dimer **3** (entry 4). At -40 °C, carbazole **4** was the sole product (entry 5). We concluded that the condition proceeded with PIFA at 0 °C is an optimized condition (entry 2).

Table 1. Optimization of reaction conditions

entry	1	R	reagent	yield (%)				
				2	3	4	1	
1	1a	PMB	DMP	2a	-	-	18	59
2	"	"	PIDA	"	-	trace	-	95
3	"	"	Ph(OH)OTs	"	trace	22	5	10
4	"	"	FPIFA	"	8	29	2	5
5	"	"	PIFA	"	35	27	4	-
6 ^{a)}	1b	H	"					
7	1c	Boc	"	2c	3	-	-	28
8	1d	Ac	"	2d	9	-	-	32

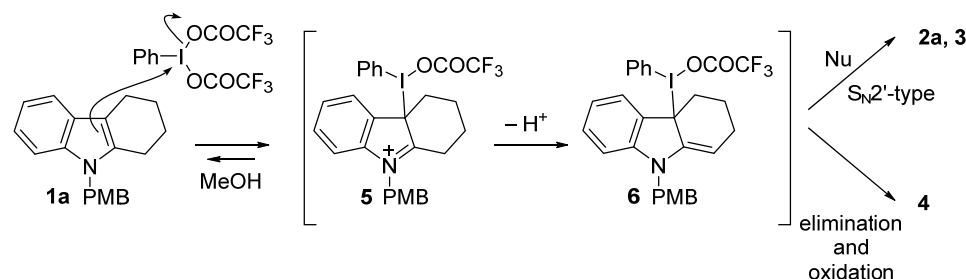
a) The reaction gave a complex-mixture.

Table 2. Effect of reaction temperature using PIFA

entry	temp. ($^\circ\text{C}$)	time (min)	yield (%)			
			2a	3	4	1a
1	r.t.	15	-	43	5	13
2	0	10	35	27	4	-
3	-20	10	30	20	3	-
4	-30	50	24	-	10	22
5	-40	60	-	-	22	33

The plausible reaction mechanism is as follows (Scheme 2). First, the 3-position of **1a** attacks an iodine atom of PIFA to generate iminium intermediate **5**. Transformation of iminium **5** to enamine **6**, followed by

the nucleophilic attack of MeOH afforded 2 α -methoxy compound **2a**. At room temperature (Table 2, entry 1), the immediate attack of unreacted **1a** to intermediate **6** produced dimer **3** before addition of MeOH. Since **1a** was consumed at 0 °C and –20 °C (checked by TLC), the regeneration of **1a** would be caused by the attack of MeOH on the iodine atom in iminium **5**. The addition of MeOH to **6** is faster than **1a**, therefore, both yield of **2a** and ratio of **2a/3** were increased (entries 2 and 3). At –30 °C and –40 °C, starting material **1a** was not consumed completely and nucleophiles (**1a** and/or MeOH) react more slowly with **6**. Consequently, the elimination of iodobenzene from intermediates and the subsequent oxidation occurs to give carbazole **4** (entries 4 and 5). The structures of **2a-d** and **3** were determined by differential nOe and H-H cosy spectra (Figure 1).



Scheme 2. Plausible reaction mechanism

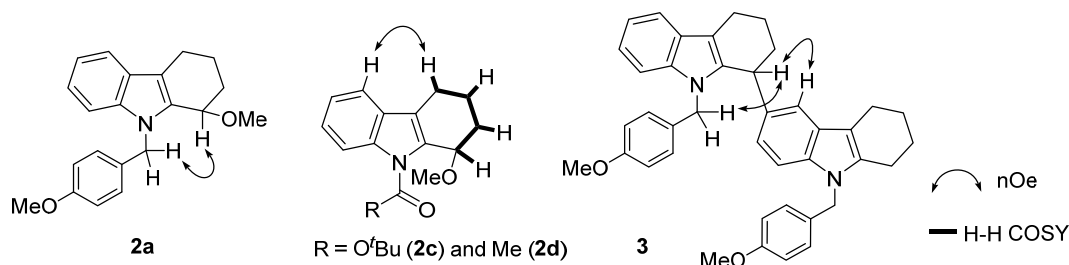
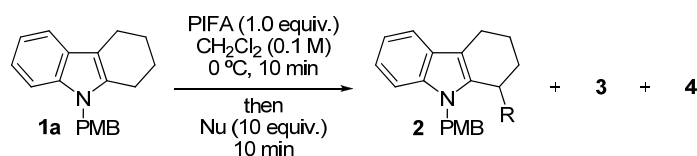


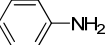
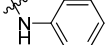


Figure 1. Structure determination of **2a-d** and **3**

Study of Nucleophiles

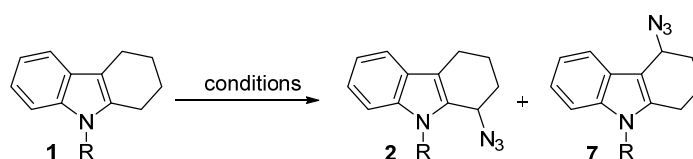
With the optimized conditions in hand, we investigated the scope and limitation of nucleophiles for this reaction (Table 3). As is the case in MeOH, an isopropoxy group was introduced to give **2e** in 41% yield (entry 1). The carbon nucleophiles MeMgBr and Me₂Zn gave 2 α -methyl derivative **2f** in 26% and 21% yields, respectively (entries 2, 3). Vinyl and allyl groups were also introduced at the 2 α -position to give products **2g** (39%) and **2h** (31%) (entries 4, 5). Additionally, the use of *N*-methylindole as an aryl nucleophile provided compound **2i** in 9% yield (entry 6). Also, the nitrogen nucleophiles, aniline and benzylamine afforded the products **2j** and **2k** in 5% and 8% yields, respectively (entries 7, 8).

Table 3. Study of nucleophiles

entry	Nu	R		yield (%)			
				2	3	4	1a
1	<i>i</i> PrOH	<i>O</i> Pr	2e	41	-	10	13
2	MeMgBr	Me	2f	26	23	-	-
3	Me ₂ Zn	"	"	21	41	-	-
4	vinylMgBr	vinyl	2g	39	8	1	15
5	allylMgBr	allyl	2h	31	-	1	17
6			2i	9	18	-	54
7			2j	5	29	1	9
8	BnNH ₂	NHBn	2k	8	-	-	64

3 α -Azide Substituent

When TMSN₃ was used as a nucleophile at 0 °C for substrate **1a**, 2 α -azide product **2l** was obtained in 20% yield (Table 4, entry 1, Method A). At -20 °C, 3 α -azide product **7a** was produced in 45% yield along with 2 α -product **2l** in 26% yield (entry 2). For *N*-Boc derivative **1c** under this reaction condition, both 2 α - and 3 α -substituted products **2m** and **7c** were obtained in 22% and 25% yields, respectively (entry 3).

Table 4. Azidation of **1** with the combination of hypervalent iodine and TMSN₃

entry	1	R	reagents	temp. (°C)	yields (%)		
					2	7	1
1	1a	PMB	PIFA, TMSN ₃ ^{a)}	0	2l 20	7a trace	18
2	"	"	"	-20	" 26	" 45	-
3	1c	Boc	"	"	2m 22	7c 25	11
4	1a	PMB	PIFA-TMSN ₃ ^{b)}	-40	2l 9	7a 33	12
5	"	"	PhIO-TMSN ₃ ^{c)}	"	" 9	" 43	-
6	1c	Boc	"	"	2m 21	7c 30	5
7	1e	CO ₂ Me	"	-40 to -25	2n 13	7e 25	4
8	1d	Ac	"	"	2o 6	7d 16	12

a) Method A: mixture of **1** and PIFA (1.0 equiv.) then addition of TMSN₃ (10 equiv.) in CH₂Cl₂ (0.1 M).

b) Method B: preformed PhI(N₃)₂ by PIFA (1.2 equiv.) and TMSN₃ (2.4 equiv.) in MeCN (0.1 M).

c) Method B: preformed PhI(N₃)₂ by PhIO (1.2 equiv.) and TMSN₃ (2.4 equiv.) in CH₂Cl₂ (0.1 M).

Since the treatment of **1c** with MeOH produced trace of 2 α -derivative **2c** (3% yield, Table 1, entry 7), we hypothesized that the mechanism of azidation is different from that of other nucleophiles as shown in Table 3. It is known that combinations of iodosobenzene (PhIO) and TMSN₃ generate phenyliodine bisazide (PhI(N₃)₂),¹² which is an extremely labile intermediate for producing radical species.¹³ Thus, we investigated the use of preformed PhI(N₃)₂ for the azidation¹⁴ of **1** (Table 4, entries 4-8, Method B). To a solution of PIFA in MeCN was added TMSN₃ at -40 °C and the mixture was stirred for 5 min. After the addition of **1a**, 2 α -azide **2l** and 3 α -azide **7a** were formed in 9% and 33% yields, respectively (entry 4). The combination of PhIO and TMSN₃ increased the yield of **7a** to 43% (entry 5). The other substrates **1c-e** gave 3 α -azide compounds **7c-e** in 16-30 % yields (entries 6-8). The position of the substituted azide group in **2** and **7** was determined by differential nOe and H-H cosy spectra (Figure 2).

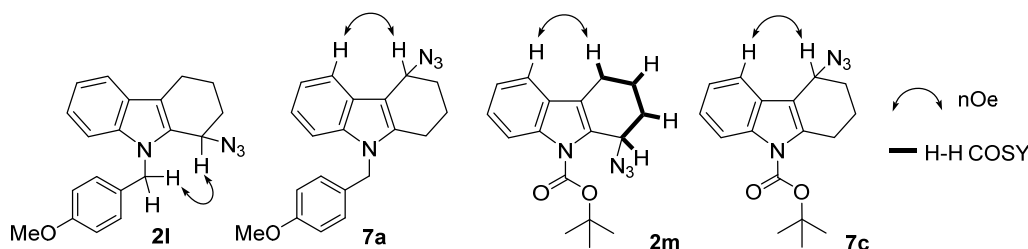
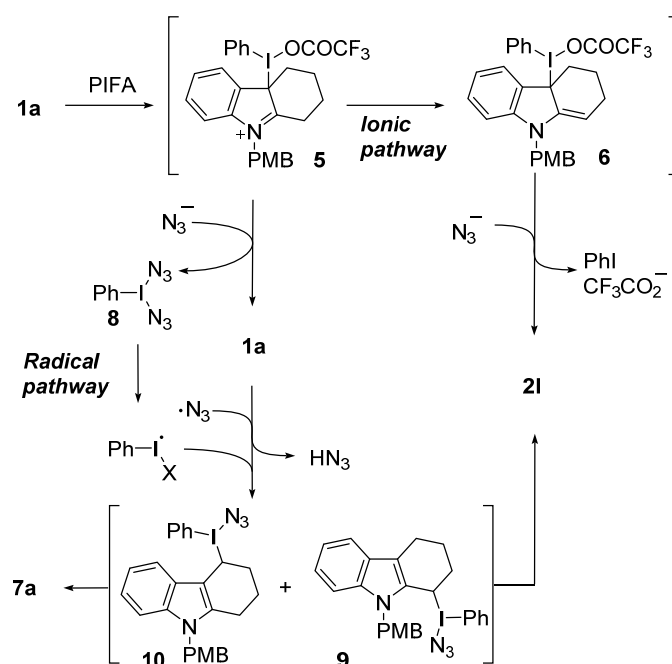


Figure 2. Structure determination of **2l,m** and **7a,c**

Based on these results, we suggest the following plausible mechanism included both ionic and radical pathways (Scheme 3). In the ionic pathway, **1a** and PIFA generate iminium **5** followed by formation of enamine **6** and then S_N2'-type reaction between the azide ion and **6** gives 2 α -compound **2l**. In the radical



Scheme 3. Plausible reaction mechanism

pathway, the iodine atom of iminium **5** or PIFA reacts with TMSN_3 to produce $\text{PhI}(\text{N}_3)_2$ **8**. Then, the azide radical from **8** reacts with tetrahydrocarbazole **1a** at 2α and/or 3α position and then the generated benzylic radical coupled with the iodo radical to form intermediates **9** and **10**. Finally, reductive elimination of iodobenzene gave the corresponding products **2l** and **7a**.

Using the combination of iodobenzene and TMSN_3 (Table 4, entry 5), we studied the scope and limitations of the substrates for a variety of 2,3-disubstituted indoles (Table 5). *N*-Boc cyclopenta[*b*]indole **1g** and cyclohepta[*b*]indoles **1i** produced the corresponding 2α - and 3α -azide products **2** and **7** in higher yield than *N*-PMB derivatives **1f** and **1h** (entries 1 vs 2, 3 vs 4). The reaction of cyclohepta[*b*]indole **1h** was accompanied by formation of olefin **11** in 23% yield (entry 3). In the case of 2,3-dialkyl substituted indoles, *N*-PMB derivatives gave better yields and selectivity than the *N*-Boc indoles (entries 5 vs 6, 7 vs 8). In particular, *N*-PMB-2-propyl-3-ethylindole **1n** gave only 3α -product **7n** in 75% yield (entry 9).

Table 5. Study of various 2,3-disubstituted indole derivatives

1 R = PMB or Boc

entry	1	yield (%)		entry	1	yield (%)			
		2	7			2	7		
1		1f : R = PMB	2p 8	7f N.A. ^{a)}	5		1j : R = PMB	2t 5	7j 50
2		1g : R = Boc	2q 13	7g 16	6		1k : R = Boc	2u 2	7k 11
3 ^{b)}		1h : R = PMB	2r 3	7h N.A.	7		1l : R = PMB	2v 1	7l 63
4		1i : R = Boc	2s 7	7i 18	8		1m : R = Boc	2w N.A.	7m 16
		11 PMB			9		1n : R = PMB	2x N.A.	7n 75

a) N.A. means not available.

b) Compound **11** was obtained in 23 % yield.

CONCLUSION

We studied the one-pot functionalization of 2,3-substituted indoles with hypervalent iodine reagents. We developed the substitution at 2α -position of indoles using PIFA with oxygen and carbon nucleophiles, which afforded 2α -derivatives in moderate yields. In the case of the combination of PhIO and TMSN_3 , the

radical mechanism was also included to afford 3 α -azide derivatives preferentially. These results provide an interesting complementary approach to our previous method using thionium species, which afforded 2 α -azide derivatives.

EXPERIMENTAL

All melting points were measured on a Yanagimoto micro melting point apparatus, and are uncorrected. IR spectra were recorded on a Shimadzu IR Prestige-21 spectrophotometer. ^1H and ^{13}C NMR spectra were measured on a JEOL JNM-AL300 (300 MHz), a JEOL JNM-AL400 (400 MHz), a JEOL JNM-ECS400 (400 MHz), or a JEOL JNM-LA500 (500 MHz) spectrometer with tetramethylsilane as an internal standard. J -Values are given in Hertz. Mass spectra were recorded on a JEOL JMS 700 instrument with a direct inlet system. Thin layer chromatography (TLC) was carried out on a Merck silica gel plate 60F₂₅₄. Column chromatography was carried out on a silica gel [Fuji Silysia Co. Inc. (silica gel PSQ 60B)]. All solvents were purified by standard procedures prior to use. The following compounds were characterized by the previous reports: **1a-d**, **1f-m**, **2a**, **2d-m**, **3**, **4**.^{6,8,15}

Representative procedure for C-H functionalization of indole derivatives with hypervalent iodine reagent (Table 1-3, Table 4 entries 1-3)

Under argon atmosphere, to a solution of **1a** (92 mg, 0.31 mmol) in CH_2Cl_2 (0.1 M) was added PIFA (0.13 g, 0.31 mmol) at 0 °C. After 10 min stirring, MeOH (0.13 mL, 3.1 mmol) was added to the reaction mixture. The mixture was stirred for 10 min and quenched by aqueous sodium sulfite (3 mL) and then extracted with CH_2Cl_2 (10 mL, 3 times). The organic layer was washed by brine and dried over MgSO_4 . The concentrated residue was purified by silica gel chromatography (AcOEt/*n*-hexane = 1/20) to give **2a**⁸ (34 mg, 35%); **3**⁸ (24 mg, 27%); **4**¹⁵ (3.5 mg, 4%).

9-(*tert*-Butoxycarbonyl)-1-methoxy-1,2,3,4-tetrahydro-9H-carbazole (**2c**)

Colorless oil. IR (CHCl_3): 2982, 2934, 1722, 1454, 1371, 1314 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ 1.69 (9H, s, *t*Bu), 1.78-2.04 (3H, m, $\text{CH}_2\text{CH}_2\text{CHOMe}$, $\text{CH}_2\text{CH}_2\text{CHOMe}$), 2.28 (1H, ddt, $J = 3.0, 6.6, 13.8$ Hz, CH_2CHOMe), 2.53 (1H, ddd, $J = 6.3, 11.4, 16.8$ Hz, CCH_2CH_2), 2.80 (1H, ddd, $J = 2.1, 5.4, 16.2$ Hz, CCH_2CH_2), 3.50 (3H, s, OCH_3), 5.04 (1H, t, $J = 3.0$ Hz, CHOMe), 7.19 (1H, dt, $J = 0.9, 7.2$ Hz, Ar-H), 7.27 (1H, dt, $J = 1.5, 7.2$ Hz, Ar-H), 7.43 (1H, dd, $J = 0.6, 8.1$ Hz, Ar-H), 8.02 (1H, d, $J = 8.1$ Hz, Ar-H). ^{13}C NMR (75 MHz, CDCl_3): δ 17.2, 21.1, 27.1, 28.3, 56.3, 70.6, 83.2, 115.7, 118.5, 119.8, 122.2, 128.9, 124.5, 134.2, 136.2, 150.3. MS (EI): m/z (%) 301 (M^+ , 34), 214 (11), 201 (54), 170 (63), 169 (100), 168 (44), 167 (13), 57 (29). HRMS (EI): m/z Calcd for $\text{C}_{18}\text{H}_{23}\text{NO}_3$: 301.1678; Found: 301.1680.

9-(4-Methoxybenzyl)-1-phenylamino-1,2,3,4-tetrahydro-9H-carbazole (2j)

Yellow oil. IR (CHCl₃): 2928, 2839, 1601, 1512, 1501, 1464, 1246 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.77-1.91 (3H, m, CH₂CH₂CHN, CH₂CH₂CHNHAr), 2.17-2.27 (1H, m, CH₂CHNHAr), 2.67 (1H, dt, *J* = 5.4, 8.7 Hz, CCH₂CH₂), 2.90 (1H, dt, *J* = 4.2, 16.2 Hz, CCH₂CH₂), 3.75 (3H, s, OCH₃), 3.83 (1H, brs, NHPh), 4.65 (1H, brs, CHNHAr), 5.20 (1H, d, *J* = 16.8 Hz, NCH₂Ar), 5.31 (1H, d, *J* = 16.8 Hz, NCH₂Ar), 6.53 (2H, d, *J* = 7.8 Hz, Ar-H), 6.67-6.71 (1H, m, Ar-H), 6.77-6.73 (2H, m, Ar-H), 6.84-6.86 (2H, m, Ar-H), 7.11 (1H, ddd, *J* = 0.9, 6.6, 7.5 Hz, Ar-H), 7.17 (2H, dd, *J* = 7.5, 8.4 Hz, Ar-H), 7.18 (1H, dt, *J* = 1.2, 6.6 Hz, Ar-H), 7.26 (1H, dt, *J* = 0.9, 7.2 Hz, Ar-H), 7.57 (1H, dt, *J* = 0.8, 7.5 Hz, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 18.4, 21.0, 28.5, 44.9, 45.8, 55.2, 109.5, 112.6, 113.0, 114.0, 117.4, 118.6, 119.0, 122.1, 126.6, 127.3, 129.3, 130.4, 134.2, 137.1, 146.3, 158.7. MS (EI) *m/z* (%): 382 (M⁺, 1), 290 (23), 289 (56), 121 (100), 93 (13). HRMS (EI): *m/z* Calcd for C₂₆H₂₆N₂O: 382.2045; Found: 382.2040.

Representative procedure for C-H functionalization of indole derivatives with hypervalent iodine reagent (Table 4 entries 4-8, Table 5)

Under argon atmosphere, a solution of iodosobenzene (PhIO, 57 mg, 0.26 mmol) in CH₂Cl₂ (2.2 mL) was added trimethylsilyl azide (TMSN₃, 72 μL, 0.52 mmol) and stirred for 5 min. Subsequently, **1a** (63 mg, 0.22 mmol) was added to the reaction mixture and stirred for 60 min at -40 °C. The mixture was quenched by aqueous sodium sulfite (2 mL) and then extracted with CH₂Cl₂ (10 mL, 3 times). The organic layer was washed by brine and dried by MgSO₄. The concentrated residue was purified by silica gel chromatography (AcOEt/*n*-hexane = 1/20) or preparative silica gel chromatography to give **2l** (6.4 mg, 9%); **7a** (32 mg, 43%).

4-Azido-9-(4-methoxybenzyl)-1,2,3,4-tetrahydro-9H-carbazole (7a)

Brown oil. IR (CHCl₃): 2947, 2837, 2097, 1730, 1612, 1512, 1464 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.88-2.15 (4H, m, CHN₃CH₂CH₂, CHN₃CH₂CH₂), 2.53-2.63 (1H, m, CCH₂), 2.73 (1H, dt, *J* = 4.8, 16.2 Hz, CCH₂), 3.73 (3H, s, OCH₃), 4.88 (1H, t, *J* = 4.2 Hz, CHN₃), 5.19 (2H, d, *J* = 1.8 Hz, NCH₂Ar), 6.78-6.83 (2H, m, Ar-H), 6.90-6.95 (2H, m, Ar-H), 7.12-7.20 (2H, m, Ar-H), 7.23-7.29 (1H, m, Ar-H), 7.68-7.73 (1H, m, Ar-H). ¹³C NMR (75 MHz, CDCl₃): δ 19.0, 21.9, 29.9, 45.8, 54.9, 55.2, 107.5, 109.4, 114.1, 118.2, 119.9, 121.6, 126.6, 127.3, 129.4, 136.6, 138.2, 158.8. HRMS (FAB): *m/z* Calcd for C₂₀H₂₀N₄O: 332.1637; Found: 332.1635.

4-Azido-9-(tert-butoxycarbonyl)-1,2,3,4-tetrahydro-9H-carbazole (7c)

Yellow oil. IR (CHCl₃): 3009, 2980, 2945, 2934, 2100, 1724, 1454, 1369, 1314 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.71 (9H, s, ^tBu), 1.93-2.12 (4H, m, CHN₃CH₂CH₂, CHN₃CH₂CH₂), 2.84-3.04 (1H, m, CCH₂),

3.18 (1H, dt, $J = 4.8, 18.0$ Hz, CCH₂), 4.72 (1H, t, $J = 3.9$ Hz, CHN₃), 7.22-7.32 (2H, m, Ar-H), 7.57-7.67 (1H, m, Ar-H), 8.08-8.18 (1H, m, Ar-H).

¹³C NMR (75 MHz, CDCl₃): δ 19.5, 25.6, 28.2, 28.9, 54.2, 84.0, 114.1, 115.6, 118.0, 123.0, 124.0, 128.2, 135.8, 138.7, 150.4. MS (EI): m/z (%) 312 (19), 270 (16), 215 (14), 214 (100), 213 (19), 170 (66), 169 (20), 168 (30), 167 (12), 57 (41). HRMS (EI): m/z Calcd for C₁₇H₂₀N₄O₂: 312.1586; Found: 312.1585.

1-Azido-9-(methoxycarbonyl)-1,2,3,4-tetrahydro-9H-carbazole (2n)

White solid. mp 93-95 °C. IR (CHCl₃): 2953, 2100, 1734, 1456, 1443, 1369 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.89-2.05 (3H, m, CH₂CH₂CHN₃), 2.15-2.29 (1H, m, CH₂CH₂CHN₃), 2.50-2.65 (1H, m, CCH₂), 2.85 (1H, dt, $J = 3.6, 16.8$ Hz, CCH₂), 4.10 (3H, s, OCH₃), 5.29 (1H, t, $J = 3.0$ Hz, CHN₃), 7.27 (1H, dt, $J = 0.9, 7.5$ Hz, Ar-H), 7.36 (1H, dt, $J = 1.2, 7.2$ Hz, Ar-H), 7.47 (1H, dt, $J = 0.6, 7.8$ Hz, Ar-H), 8.15 (1H, d, $J = 8.1$ Hz, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 17.4, 20.8, 30.7, 53.8, 55.1, 116.0, 118.8, 121.3, 123.1, 125.5, 128.8, 131.4, 136.1, 152.0. MS (EI): m/z (%) 270 (M⁺, 15), 229 (15), 228 (100), 168 (16), 167 (11). HRMS (EI): m/z Calcd for C₁₄H₁₄N₄O₂: 270.1117; Found: 270.1115.

4-Azido-9-(methoxycarbonyl)-1,2,3,4-tetrahydro-9H-carbazole (7e)

Yellowish white solid. mp 77-79 °C. IR (CHCl₃): 2955, 2359, 2340, 2099, 1734, 1458, 1443, 1368 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.90-2.17 (4H, m, CHN₃CH₂CH₂), 2.80-3.04 (1H, m), 3.20 (1H, dt, $J = 4.5, 18.3$ Hz, CCH₂), 4.05 (3H, s, OCH₃), 4.72 (1H, t, $J = 4.2$ Hz, CHN₃), 7.30 (1H, t, $J = 3.3$ Hz, Ar-H), 7.30 (1H, ddd, $J = 1.8, 7.2, 17.1$ Hz, Ar-H), 7.56-7.67 (1H, m, Ar-H), 8.07-8.20 (1H, m, Ar-H). ¹³C NMR (100 MHz, CDCl₃): 19.4, 25.2, 28.9, 53.5, 54.1, 114.8, 115.6, 118.2, 123.3, 124.3, 128.3, 135.7, 138.6, 152.4. MS (EI): m/z (%) 270 (M⁺, 12), 229 (15), 228 (100), 168 (16), 167 (12). HRMS (EI): m/z Calcd for C₁₄H₁₄N₄O₂: 270.1117; Found: 270.1118.

9-Acetyl-1-azido-1,2,3,4-tetrahydro-9H-carbazole (2o)

Brown oil. IR (CHCl₃): 2947, 2930, 2100, 1697, 1460, 1373, 1308 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.91-2.05 (3H, m, CH₂CH₂CHN₃), 2.12-2.25 (1H, m, CH₂CH₂CHN₃), 2.504-2.70 (1H, m, CCH₂), 2.84 (3H, s, COCH₃), 2.81-2.91 (1H, m, CCH₂), 5.39 (1H, t, $J = 3.0$ Hz, CHN₃), 7.29 (1H, dt, $J = 0.9, 7.5$ Hz, Ar-H), 7.37 (1H, dt, $J = 1.2, 7.2$ Hz, Ar-H), 7.51 (1H, dt, $J = 0.6, 7.8$ Hz, Ar-H), 7.76 (1H, dt, $J = 0.6, 8.1$ Hz, Ar-H). ¹³C NMR (75 MHz, CDCl₃): δ 17.4, 20.8, 27.4, 30.7, 55.3, 114.7, 119.5, 121.8, 123.1, 125.3, 129.5, 132.7, 135.7, 169.6. MS (EI): m/z (%) 254 (M⁺, 25), 213 (11), 212 (74), 211 (17), 184 (18), 183 (17), 171 (13), 170 (100), 169 (38), 168 (56), 167 (23), 156 (12). HRMS (EI): m/z Calcd for C₁₄H₁₄N₄O: 254.1168; Found: 254.1167.

9-Acetyl-4-azido-1,2,3,4-tetrahydro-9H-carbazole (7d)

Colorless solid. mp 76-79 °C. IR (CHCl₃): 3009, 2949, 2099, 1701, 1371, 1304 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.90-2.30 (4H, m, CHN₃CH₂CH₂), 2.74 (3H, s, COCH₃), 2.91-3.04 (1H, m, CCH₂), 3.17 (1H, dt, *J* = 4.5, 17.7 Hz, CCH₂), 4.73 (1H, t, *J* = 3.9 Hz, CHN₃), 7.31 (1H, t, *J* = 3.6 Hz, Ar-H), 7.31 (1H, ddd, *J* = 2.1, 7.2, 15.6 Hz, Ar-H), 7.60-7.70 (1H, m, Ar-H), 7.92-8.02 (1H, m, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 19.8, 26.3, 27.4, 28.7, 54.2, 115.2, 115.5, 118.6, 123.5, 124.5, 128.8, 135.7, 138.7, 170.0. MS (EI): *m/z* (%) 254 (M⁺, 13), 212 (43), 211 (15), 184 (11), 171 (12), 170 (100), 169 (33), 168 (46), 167 (20), 156 (13). HRMS (EI): *m/z* Calcd for C₁₄H₁₄N₄O: 254.1168; Found: 254.1161.

3-Azido-4-(4-methoxybenzyl)-1,2,3,4-tetrahydrocyclopenta[b]indole (2p)

Brown oil. IR (CHCl₃): 3007, 2936, 2864, 2094, 1512, 1464, 1246 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ: 2.62 (1H, ddt, *J* = 2.8, 7.6, 16.4 Hz, CH₂CH₂CHN₃), 2.82 (1H, ddd, *J* = 3.6, 8.8, 14.0 Hz, CH₂CH₂CHN₃), 2.95 (1H, ddt, *J* = 6.0, 8.0, 13.8 Hz, CH₂CH₂CHN₃), 3.02-3.15 (1H, m, CH₂CH₂CHN₃), 3.77 (3H, s, OCH₃), 4.64 (1H, dt, *J* = 2.8, 9.2 Hz, CHN₃), 5.19 (1H, d, *J* = 16.1 Hz, NCH₂Ar), 5.34 (1H, d, *J* = 15.6 Hz, NCH₂Ar), 6.77-6.89 (2H, m, Ar-H), 7.03-7.08 (2H, m, Ar-H), 7.10 (1H, td, *J* = 1.0, 8.0 Hz, Ar-H), 7.17 (1H, td, *J* = 1.6, 8.4 Hz, Ar-H), 7.25 (1H, td, *J* = 1.2, 8.4 Hz, Ar-H), 7.52 (1H, dt, *J* = 0.8, 7.6 Hz, Ar-H). ¹³C NMR (100 MHz, CDCl₃): 23.2, 36.7, 47.6, 55.2, 60.0, 110.5, 114.1, 119.6, 119.8, 121.9, 122.2, 123.5, 127.9, 129.6, 141.2, 142.0, 159.0. MS (EI): *m/z* (%) 318 (M⁺, 12), 290 (13), 276 (29), 275 (23), 121 (100). HRMS (EI): *m/z* Calcd for C₁₉H₁₈N₄O: 318.1481; Found: 318.1479.

3-Azido-4-(tert-butoxycarbonyl)-1,2,3,4-tetrahydrocyclopenta[b]indole (2q)

Yellowish green oil. IR (CHCl₃): 2980, 2936, 2099, 1730, 1368, 1319 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.69 (9H, s, ^tBu), 2.56 (1H, ddd, *J* = 0.8, 6.0, 9.9 Hz, CH₂CH₂CHN₃), 2.70-2.92 (2H, m, CH₂CH₂CHN₃), 2.98 (1H, ddt, *J* = 1.5, 2.1, 12.3 Hz, CH₂CH₂CHN₃), 5.10 (1H, d, *J* = 6.6 Hz, CHN₃), 7.25 (1H, dt, *J* = 1.2, 7.5 Hz, Ar-H), 7.33 (1H, dt, *J* = 1.2, 7.2 Hz, Ar-H), 7.46 (1H, dd, *J* = 0.6, 5.7 Hz, Ar-H), 8.20 (1H, d, *J* = 8.1 Hz, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 22.8, 28.2, 36.6, 62.0, 84.1, 116.2, 119.8, 123.0, 124.9, 125.5, 129.1, 139.8, 140.8, 149.3. MS (EI): *m/z* (%) 298 (M⁺, 26), 256 (20), 242 (11), 201 (13), 200 (97), 169 (21), 157 (12), 156 (100), 155 (47), 57 (43). HRMS (EI): *m/z* Calcd for C₁₆H₁₈N₄O₂: 298.1430; Found: 298.1424.

1-Azido-4-(tert-butoxycarbonyl)-1,2,3,4-tetrahydrocyclopenta[b]indole (7g)

Brown oil. IR (CHCl₃): 3007, 2980, 2936, 2093, 1730, 1369, 1321 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.66 (9H, s, ^tBu), 2.49 (1H, ddt, *J* = 3.0, 8.4, 14.4 Hz, CHN₃CH₂CH₂), 2.81-2.98 (1H, m, CHN₃CH₂CH₂), 3.06 (1H, ddd, *J* = 3.6, 8.7, 17.1 Hz, CHN₃CH₂CH₂), 3.27 (1H, dddd, *J* = 1.8, 5.1, 7.8, 16.8 Hz, CHN₃CH₂CH₂), 4.96 (1H, dt, *J* = 2.1, 7.8 Hz, CHN₃), 7.27 (1H, ddd, *J* = 1.8, 7.2, 19.2 Hz, Ar-H), 7.27 (1H,

t, $J = 2.1$ Hz, Ar-H), 7.52 (1H, dd, $J = 2.7, 6.9$ Hz, Ar-H), 8.18 (1H, dd, $J = 1.8, 6.6$ Hz, Ar-H). ^{13}C NMR (100 MHz, CDCl_3): δ 27.9, 28.2, 35.6, 60.7, 83.9, 115.9, 118.7, 122.5, 123.2, 123.8, 125.1, 140.3, 146.3, 149.6. MS (EI): m/z (%) 298 (M^+ , 14), 256 (22), 201 (13), 200 (100), 199 (16), 169 (11), 156 (56), 155 (22), 154 (12), 57 (41). HRMS (EI): m/z $\text{C}_{16}\text{H}_{18}\text{N}_4\text{O}_2$: 298.1430; Found: 298.1424.

6-Azido-5-(4-methoxybenzyl)-5,6,7,8,9,10-hexahydrocyclohepta[b]indole (2r)

Brown oil. IR (CHCl_3): 2930, 2100, 1512, 1464 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ 1.50-1.75 (1H, m, CCH_2CH_2), 1.83 (1H, ddt, $J = 3.0, 12.0, 16.8$ Hz, CH_2CHN_3), 1.90-2.12 (3H, m, $\text{CCH}_2\text{CH}_2\text{CH}_2$), 2.20 (1H, ddt, $J = 2.4, 5.1, 13.2$ Hz, CH_2CHN_3), 2.86 (1H, ddd, $J = 2.7, 11.7, 14.7$ Hz, CCH_2), 3.10 (1H, ddd, $J = 2.4, 6.0, 15.9$ Hz, CCH_2), 3.76 (3H, s, OCH_3), 4.76 (1H, dd, $J = 2.7, 5.4$ Hz, CHN_3), 5.32 (1H, d, $J = 17.1$ Hz, NCH_2Ar), 5.41 (1H, d, $J = 17.1$ Hz, NCH_2Ar), 6.76-6.85 (2H, m, Ar-H), 6.85-6.95 (2H, m, Ar-H), 7.14 (1H, ddd, $J = 1.5, 6.9, 8.1$ Hz, Ar-H), 7.21 (1H, dt, $J = 1.2, 6.6$ Hz, Ar-H), 7.27 (1H, d, $J = 7.2$ Hz, Ar-H), 7.61 (1H, d, $J = 6.9$ Hz, Ar-H). ^{13}C NMR (75 MHz, CDCl_3): δ 23.7, 24.8, 27.9, 32.3, 46.0, 55.2, 57.1, 109.6, 114.2, 117.5, 119.0, 119.4, 122.6, 126.9, 127.3, 129.8, 132.9, 136.4, 158.9. MS (EI): m/z (%) 346 (M^+ , 3), 304 (21), 303 (57), 121 (100). HRMS (EI): m/z Calcd for $\text{C}_{21}\text{H}_{22}\text{N}_4\text{O}$: 346.1794; Found: 346.1791.

5-(4-Methoxybenzyl)-5,6,7,8-tetrahydrocyclohepta[b]indole (11)

Green oil. IR (CHCl_3): 2932, 1612, 1512, 1468, 1248 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ 2.02 (2H, quint, $J = 5.7$ Hz, $\text{CH}_2\text{CH}_2\text{CH}_2$), 2.46 (2H, dd, $J = 5.1, 10.8$ Hz, CHCH_2CH_2), 2.95 (2H, t, $J = 5.7$ Hz, CCH_2CH_2), 3.74 (3H, s, OCH_3), 5.24 (2H, s, NCH_2Ar), 5.74 (1H, dt, $J = 5.7, 11.4$ Hz, CHCH_2CH_2), 6.68 (1H, dt, $J = 1.5, 11.4$ Hz, CHCH_2CH_2), 6.73-6.84 (2H, m, Ar-H), 6.85-7.00 (2H, m, Ar-H), 7.08-7.18 (2H, m, Ar-H), 7.18-7.22 (1H, m, Ar-H), 7.60-7.70 (1H, m, Ar-H). ^{13}C NMR (100 MHz, CDCl_3): δ 23.3, 28.3, 30.6, 45.9, 55.2, 109.0, 110.8, 114.2, 117.6, 119.6, 120.0, 121.3, 125.3, 127.2, 127.5, 129.7, 136.3, 138.6, 158.8. MS (EI): m/z (%) 303 (M^+ , 44), 121 (100). HRMS (EI): m/z Calcd for $\text{C}_{21}\text{H}_{21}\text{NO}$: 303.1623; Found: 303.1622.

6-Azido-5-(tert-butoxycarbonyl)-5,6,7,8,9,10-hexahydrocyclohepta[b]indole (2s)

Dark green oil. IR (CHCl_3): 2932, 2104, 1722, 1454, 1371, 1360, 1315, 1308 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ 1.71 (9H, s, ^tBu), 1.45-1.81 (1H, m, $\text{CHN}_3\text{CH}_2\text{CH}_2\text{CH}_2$), 1.81-1.98 (2H, m, $\text{CHN}_3\text{CH}_2\text{CH}_2$), 1.98-2.13 (2H, m, $\text{CHN}_3\text{CH}_2\text{CH}_2\text{CH}_2$), 2.13-2.23 (1H, m, CHN_3CH_2), 2.75-2.97 (2H, m, CCH_2), 5.92 (1H, dd, $J = 2.1, 6.3$ Hz, CHN_3), 7.24 (1H, dt, $J = 1.2, 7.5$ Hz, Ar-H), 7.31 (1H, dt, $J = 1.5, 7.2$ Hz, Ar-H), 7.50 (1H, dt, $J = 0.6, 7.5$ Hz, Ar-H), 8.04 (1H, dt, $J = 1.2-7.5$ Hz, Ar-H). ^{13}C NMR (100 MHz, CDCl_3): δ 23.1, 24.4, 26.6, 28.3, 31.5, 57.4, 84.5, 115.8, 118.7, 122.6, 124.8, 124.9, 129.3, 134.4, 135.5, 150.6. MS (EI): m/z (%) 326 (M^+ , 20), 228 (34), 227 (30), 226 (16), 185 (15), 184 (100), 183 (30), 182 (32), 180 (30), 169 (15), 168 (14), 57 (42). HRMS (EI): m/z Calcd for $\text{C}_{18}\text{H}_{22}\text{N}_4\text{O}_2$: 326.1743; Found: 326.1746.

10-Azido-5-(*tert*-butoxycarbonyl)-5,6,7,8,9,10-hexahydrocyclohepta[*b*]indole (7i)

Dark green oil. IR (CHCl₃): 2981, 2932, 2102, 1726, 1456, 1371, 1354, 1312 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.69 (9H, s, ^tBu), 1.60-1.80 (1H, m, CHN₃CH₂CH₂CH₂), 1.82-1.98 (2H, m, CHN₃CH₂CH₂), 1.98-2.16 (2H, m, CHN₃CH₂CH₂CH₂), 2.16-2.36 (1H, m, CHN₃CH₂), 3.15 (1H, ddd, *J* = 2.4, 9.9, 17.1 Hz, CCH₂), 3.45 (1H, ddd, *J* = 2.4, 8.1, 17.1 Hz, CCH₂), 5.10 (1H, dd, *J* = 1.8, 5.7 Hz, CHN₃), 7.20-7.30 (2H, m, Ar-H), 7.44-7.60 (1H, m, Ar-H), 7.94-8.07 (1H, m, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 24.4, 26.5, 27.3, 28.3, 31.5, 56.5, 84.2, 115.2, 117.2, 117.5, 122.8, 123.8, 129.3, 135.2, 142.3, 150.6. MS (EI): *m/z* (%) 326 (M⁺, 19), 284 (17), 283 (25), 229 (15), 228 (100), 227 (89), 185 (10), 184 (76), 183 (57), 182 (51), 180 (12), 169 (11), 168 (31), 167 (18), 57 (66). HRMS (EI): *m/z* Calcd for C₁₈H₂₂N₄O₂: 326.1743; Found: 326.1742.

2-(Azidomethyl)-1-(4-methoxybenzyl)-3-methyl-1*H*-indole (2t)

Brown oil. IR (CHCl₃): 3007, 2936, 2108, 1512, 1248 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.40 (3H, s, CCH₃), 3.75 (3H, s, OCH₃), 4.40 (2H, s, CH₂N₃), 5.33 (2H, s, NCH₂Ar), 6.74-6.83 (2H, m, Ar-H), 6.86-6.95 (2H, m, Ar-H), 7.14 (1H, ddd, *J* = 1.5, 6.3, 7.5 Hz, Ar-H), 7.22 (1H, dt, *J* = 1.5, 8.1 Hz, Ar-H), 7.24-7.29 (1H, m, Ar-H), 7.61 (1H, dt, *J* = 1.2, 7.8 Hz, Ar-H). ¹³C NMR (75 MHz, CDCl₃): δ 8.9, 44.2, 46.3, 55.2, 109.6, 111.9, 114.2, 119.3, 119.4, 122.8, 127.1, 127.8, 128.9, 129.8, 137.1, 158.9. MS (EI): *m/z* (%) 306 (M⁺, 23), 278 (28), 264, (21), 157 (15), 121 (100). HRMS (EI): *m/z* Calcd for C₁₈H₁₈N₄O: 306.1480; Found: 306.1478.

3-(Azidomethyl)-1-(4-methoxybenzyl)-2-methyl-1*H*-indole (7j)

Brown oil. IR (CHCl₃): 3007, 2936, 2106, 1512, 1248 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 2.39 (3H, s, CCH₃), 3.75 (3H, s, OCH₃), 4.55 (2H, s, CH₂N₃), 5.28 (2H, s, NCH₂Ar), 6.76-6.82 (2H, m, Ar-H), 6.86-6.93 (2H, m, Ar-H), 7.11-7.20 (2H, m, Ar-H), 7.22-7.28 (1H, m, Ar-H), 7.59-7.68 (1H, m, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 10.4, 45.4, 46.2, 55.3, 105.8, 109.4, 114.2, 117.9, 120.0, 121.7, 127.1, 127.5, 129.4, 136.0, 136.5, 158.9. MS (EI): *m/z* (%) 306 (M⁺, 12), 278 (16), 264 (27), 121 (100). HRMS (EI): *m/z* Calcd for C₁₈H₁₈N₄O: 306.1481; Found: 306.1478.

2-(Azidomethyl)-1-(*tert*-butoxycarbonyl)-3-methyl-1*H*-indole (2u)

Yellowish green oil. IR (CHCl₃): 2982, 2928, 2102, 1724, 1454, 1357, 1339, 1329 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.71 (9H, s, ^tBu), 2.32 (3H, s, CH₃), 4.78 (2H, s, CH₂N₃), 7.27 (1H, dt, *J* = 1.2, 9.6 Hz, Ar-H), 7.35 (1H, dt, *J* = 1.5, 7.2 Hz, Ar-H), 7.52 (1H, dd, *J* = 0.9, 7.5 Hz, Ar-H), 8.12 (1H, d, *J* = 9.6 Hz, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 8.8, 28.2, 45.6, 84.4, 115.9, 119.06, 119.11, 122.7, 125.3, 129.4, 129.7, 136.1, 150.2. MS (EI): *m/z* (%) 286 (M⁺, 34), 230 (13), 188 (28), 186 (26), 159 (16), 158 (32), 157 (29), 145 (13), 144 (100), 143 (24), 130 (25), 57 (80), 41 (11). HRMS (EI): *m/z* Calcd for C₁₅H₁₈N₄O₂:

286.1430; Found: 286.1428.

3-(Azidomethyl)-1-(*tert*-butoxycarbonyl)-2-methyl-1*H*-indole (7k)

Yellowish green oil. IR (CHCl₃): 2982, 2934, 2108, 1730, 1458, 1358 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.70 (9H, s, ^tBu), 2.63 (3H, s, CH₃), 4.46 (2H, s, CH₂N₃), 7.21-7.35 (2H, m, Ar-H), 7.53 (1H, ddd, *J* = 3.9, 5.2, 10.1 Hz, Ar-H), 8.12 (1H, ddd, *J* = 4.5, 5.4, 11.4 Hz, Ar-H). ¹³C NMR (75 MHz, CDCl₃): δ 14.0, 28.2, 44.5, 84.1, 112.3, 115.5, 117.7, 122.9, 124.0, 128.8, 135.6, 136.6, 150.5. MS (EI): *m/z* (%) 286 (M⁺, 47), 230 (39), 213 (11), 188 (76), 158 (25), 157 (23), 145 (12), 144 (100), 143 (17), 130 (14), 57 (89), 41 (12). HRMS (EI): *m/z* Calcd for C₁₅H₁₈N₄O₂: 286.1430; Found: 286.1427.

2-(1-Azidoethyl)-1-(4-methoxybenzyl)-3-methyl-1*H*-indole (2v)

Yellow oil. IR (CHCl₃) 3007, 2932, 2106, 1512, 1466, 1246 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.47 (3H, d, *J* = 7.2 Hz, CHN₃CH₃), 2.43 (3H, s, CCH₃), 3.75 (3H, s, OCH₃), 5.05 (1H, q, *J* = 7.2 Hz, CHN₃), 5.37 (1H, d, *J* = 17.4 Hz, NCH₂Ar), 5.46 (1H, d, *J* = 17.4 Hz, NCH₂Ar), 6.75-6.83 (2H, m, Ar-H), 6.84-6.93 (2H, m, Ar-H), 7.10-7.18 (3H, m, Ar-H), 7.59-7.63 (1H, m, Ar-H). ¹³C NMR (75 MHz, CDCl₃): δ 9.0, 20.5, 46.8, 53.8, 55.2, 109.7, 114.1, 119.0, 119.3, 122.5, 126.9, 128.2, 130.0, 133.2, 137.0, 144.7, 158.9. MS (EI): *m/z* (%) 320 (M⁺, 24), 278 (33), 277 (29), 122 (12), 121 (100). HRMS (EI): *m/z* Calcd for C₁₉H₂₀N₄O: 320.1637; Found: 320.1635.

3-(Azidomethyl)-2-ethyl-1-(4-methoxybenzyl)-1*H*-indole (7l)

Brown oil. IR (CHCl₃): 3007, 2970, 2936, 2108, 1512, 1466, 1248 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.17 (3H, t, *J* = 7.5 Hz, CH₂CH₃), 2.79 (2H, q, *J* = 7.5 Hz, CH₂CH₃), 3.73 (3H, s, OCH₃), 4.54 (2H, s, CH₂N₃), 5.29 (2H, s, NCH₂Ar), 6.74-6.81 (2H, m, Ar-H), 6.83-6.90 (2H, m, Ar-H), 7.10-7.22 (3H, m, Ar-H), 7.60-7.68 (1H, m, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 15.4, 17.9, 45.5, 46.1, 55.2, 105.1, 109.8, 114.2, 118.1, 120.1, 121.8, 127.0, 127.7, 129.6, 136.5, 141.9, 158.9. MS (EI): *m/z* (%) 320 (M⁺, 14), 278 (31), 121 (100). HRMS (EI): *m/z* Calcd for C₁₉H₂₀N₄O: 320.1637; Found: 320.1638.

3-(1-Azidomethyl)-1-(*tert*-butoxycarbonyl)-2-ethyl-1*H*-indole (7m)

Yellow oil. IR (CHCl₃): 2982, 2108, 1730, 1458, 1371, 1360, 1329 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 1.27 (3H, t, *J* = 7.2 Hz, CH₂CH₃), 1.70 (9H, s, ^tBu), 3.09 (2H, q, *J* = 7.2 Hz, CH₂CH₃), 4.45 (2H, s, CH₂N₃), 7.25 (1H, dt, *J* = 2.4, 7.8 Hz, Ar-H), 7.29 (1H, dt, *J* = 2.4, 7.2 Hz, Ar-H), 7.50-7.58 (1H, m, Ar-H), 8.12 (1H, dd, *J* = 1.8, 6.6 Hz, Ar-H). ¹³C NMR (100 MHz, CDCl₃): δ 15.4, 20.1, 28.2, 44.7, 84.2, 111.9, 115.7, 118.0, 123.0, 124.1, 128.8, 135.9, 142.5, 150.2. MS (EI) *m/z* (%): 300 (M⁺, 44), 244 (36), 216 (13), 203 (12), 202 (89), 172 (32), 171 (35), 159 (12), 158 (100), 157 (21), 156 (24), 155 (12), 144 (13), 143 (12), 57 (86), 41

(12). HRMS (EI): m/z Calcd for $C_{16}H_{20}N_4O_2$: 300.1586; Found: 300.1583.

3-Ethyl-1-(4-methoxybenzyl)-2-propyl-1H-indole (1n)

To a suspension of NaH (128 mg, 60% in mineral oil, 3.21 mmol) in dry DMF (5.0 mL) was added 3-ethyl-2-propylindole¹⁶ (400 mg, 214 mmol) at 0 °C. After stirring at room temperature for 10 min, the reaction mixture was added tetrabutylammonium iodide (79 mg, 0.214 mmol) and *p*-methoxybenzyl chloride (260 μ L, 2.57 mmol) and stirred for 30 min. The reaction was quenched by the addition of saturated aqueous NH_4Cl (10 mL) and extracted with Et_2O (30 mL, 3 times). The organic layer was dried over $MgSO_4$ and filtrate was concentrated. The residue was purified by silica gel chromatography ($AcOEt/n$ -hexane = 1/5) to afford **1n** (504 mg, 77%).

Yellow oil. IR ($CHCl_3$): 3005, 2963, 2932, 2870, 1612, 1512, 1468, 1246 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ 0.93 (3H, t, $J = 7.2$ Hz, $CH_2CH_2CH_3$), 1.23 (3H, t, $J = 7.2$ Hz, CCH_2CH_3), 1.50 (2H, ddd, $J = 7.2, 15.2, 15.2$ Hz, $CH_2CH_2CH_3$), 2.66 (2H, t, $J = 8.0$ Hz, $CH_2CH_2CH_3$), 2.76 (2H, q, $J = 7.2$ Hz, CCH_2CH_3), 3.71 (3H, s, OCH_3), 5.22 (2H, s, NCH_2Ar), 6.70-6.82 (2H, m, Ar-H), 6.82-6.92 (2H, m, Ar-H), 7.02-7.10 (2H, m, Ar-H), 7.11-7.22 (1H, m, Ar-H), 7.54-7.61 (1H, m, Ar-H). ^{13}C NMR (100 MHz, $CDCl_3$): δ 15.6, 17.5, 19.4, 25.3, 28.1, 47.5, 56.7, 110.9, 115.5 (2C), 119.8, 120.3, 122.2, 128.5, 129.3, 132.0, 137.8, 138.0, 160.2. MS (EI): m/z (%) 308 ($[M^+ + 1]$, 12), 307 (M^+ , 52), 121 (100). HRMS (EI): m/z Calcd for $C_{21}H_{25}NO$: 307.1936; Found: 307.1936.

3-(1-Azidoethyl)-1-(4-methoxybenzyl)-2-propyl-1H-indole (7n)

Yellowish green oil. IR ($CHCl_3$): 3007, 2961, 2934, 2872, 2104, 1612, 1512, 1466, 1248, 1223 cm^{-1} . 1H NMR (300 MHz, $CDCl_3$): δ 0.95 (3H, t, $J = 7.2$ Hz, $CH_2CH_2CH_3$), 1.42-1.62 (2H, m, $CH_2CH_2CH_3$), 1.69 (3H, d, $J = 6.9$ Hz, CHN_3CH_3), 2.71 (2H, t, $J = 7.8$ Hz, CCH_2), 3.73 (3H, s, OCH_3), 5.04 (1H, q, $J = 6.9$ Hz, $CCHN_3$), 5.25 (2H, s, NCH_2Ar), 6.74-6.82 (2H, m, Ar-H), 6.82-6.90 (2H, m, Ar-H), 7.06-7.21 (3H, m, Ar-H), 7.75-7.83 (1H, m, Ar-H). ^{13}C NMR (100 MHz, $CDCl_3$): δ 14.0, 21.5, 24.0, 26.6, 46.0, 55.2, 55.5, 109.8, 110.9, 114.1, 119.56, 119.64, 121.4, 125.7, 126.9, 129.6, 136.8, 138.2, 158.8. MS (EI): m/z (%) 348 (M^+ , 1), 306 (14), 305 (44), 122 (10), 121 (100). HRMS (EI): m/z Calcd for $C_{21}H_{24}N_4O$: 348.1950; Found: 348.1946.

ACKNOWLEDGEMENTS

We thank N. Eguchi, T. Koseki, and S. Yamada at the Analytical Center of our university for performing microanalysis, NMR and mass spectrometry measurements.

REFERENCES

1. Y. Kita, G. F. Koser, M. Ochiai, H. Tohma, A. Varvoglis, T. Wirth, and V. V. Zhdankin, 'Top. Curr. Chem.: Hypervalent Iodine Chemistry, Modern Developments in Organic Synthesis' Vol. 224, ed. by T. Wirth, Springer, Berlin, 2003, pp. 1-248; A. Varvoglis, *Tetrahedron*, 1997, **53**, 1179; R. M. Moriarty and O. Prakash, *Org. React.*, 1999, **54**, 273; V. V. Zhdankin and P. J. Stang, *Chem. Rev.*, 2008, **108**, 5299; L. F. Silva, Jr. and B. Olofsson, *Nat. Prod. Rep.*, 2011, **28**, 1722.
2. T. Dohi, M. Ito, N. Yamaoka, K. Morimoto, H. Fujioka, and Y. Kita, *Tetrahedron*, 2009, **65**, 10797.
3. T. Dohi, K. Morimoto, N. Takenaga, A. Goto, A. Maruyama, Y. Kiyono, H. Tohma, and Y. Kita, *J. Org. Chem.*, 2007, **72**, 109.
4. J. A. Campbell, C. A. Broka, L. Gong, K. A. M. Walker, and J.-H. Wang, *Tetrahedron Lett.*, 2004, **45**, 4073; D. Lubriks, I. Sokolovs, and E. Suna, *J. Am. Chem. Soc.*, 2012, **134**, 15436; D. Tu, L. Ma, X. Tong, X. Deng, and C. Xia, *Org. Lett.*, 2012, **14**, 4830; Q. Liu, Q. Y. Zhao, J. Liu, P. Wu, H. Yi, and A. Lei, *Chem. Commun.*, 2012, **48**, 3239.
5. N. A. Braun, J. D. Bray, and M. A. Ciufolini, *Tetrahedron Lett.*, 1999, **40**, 4985; X. Qi, H. Bao, and U. K. Tambar, *J. Am. Chem. Soc.*, 2011, **133**, 10050.
6. H. Zaimoku, T. Hatta, T. Taniguchi, and H. Ishibashi, *Org. Lett.*, 2012, **14**, 6088.
7. K. W. Fiori and J. Du Bois, *J. Am. Chem. Soc.*, 2007, **129**, 562.
8. K. Higuchi, M. Tayu, and T. Kawasaki, *Chem. Commun.*, 2011, **47**, 6728; M. Tayu, K. Higuchi, M. Inaba, and T. Kawasaki, *Org. Biomol. Chem.*, 2013, **11**, 496.
9. G. Büchi and R. E. Manning, *J. Am. Chem. Soc.*, 1966, **88**, 2532; H. Sakakibara and T. Kobayashi, *Tetrahedron*, 1966, **22**, 2475; M. Ikeda, F. Tabusa, Y. Nishimura, S. Kwon, and Y. Tamura, *Tetrahedron Lett.*, 1976, **27**, 2347; A. S. Bailey, J. B. Haxby, A. N. Hilton, J. M. Peach, and M. H. Vandrevala, *J. Chem. Soc., Perkin Trans. 1*, 1981, 382; Y. Murakami and H. Ishii, *Chem. Pharm. Bull.*, 1981, **29**, 699; Y. Tamura, M. Kirihara, M. Sasho, S. Akai, J. Sekihachi, R. Okunaka, and Y. Kita, *Chem. Commun.*, 1987, 1474; P. He and S. Zhu, *Tetrahedron*, 2005, **61**, 12398; Y. Nakano and D. W. Lupton, *Chem. Commun.*, 2014, 1757.
10. R. Di Fabio, R. Giovannini, B. Bertani, M. Borriello, A. Bozzoli, D. Donati, A. Falchi, D. Ghirlanda, C. P. Leslie, A. Pecunioso, G. Rumboldt, and S. Spada, *Bioorg. Med. Chem. Lett.*, 2006, **16**, 1749; K. S. Gudmundsson, P. R. Sebahar, L. D. Richardson, J. G. Catalano, S. D. Boggs, A. Spaltenstein, P. B. Sethna, K. W. Brown, R. Harvey, and K. R. Romines, *Bioorg. Med. Chem. Lett.*, 2009, **19**, 3489; L. Li, C. Beaulieu, M.-C. Carriere, D. Denis, G. Greig, D. Guay, G. O'Neill, R. Zamboni, and Z. Wang, *Bioorg. Med. Chem. Lett.*, 2010, **20**, 7462.
11. O. Neiland and B. Karele, *J. Org. Chem., USSR (Engl. Transl.)*, 1970, **6**, 889; G. F. Koser, A. G. Relenyi, A. N. Kalos, L. Rebrovic, and R. H. Wettach, *J. Org. Chem.*, 1982, **47**, 2487.

12. E. Zbiral and G. Nestler, *Tetrahedron*, 1970, **26**, 2945; F. Cech and E. Zbiral, *Tetrahedron*, 1975, **31**, 605.
13. P. Magnus, J. Lacour, and W. Weber, *J. Am. Chem. Soc.*, 1993, **115**, 9347; F. Fontana, F. Minisci, Y.-M. Yan, and L. Zhao, *Tetrahedron Lett.*, 1993, **34**, 2517; P. Magnus, J. Lacour, P. A. Evans, M. B. Roe, and C. Hulme, *J. Am. Chem. Soc.*, 1996, **118**, 3406; C. M. Pedersen, L. G. Marinescu, and M. Bols, *Org. Biomol. Chem.*, 2005, **3**, 816.
14. R. M. Moriarty and J. S. Khosrowshahi, *Tetrahedron Lett.*, 1986, **27**, 2809; M. Arimoto, H. Yamaguchi, E. Fujita, M. Ochiai, and Y. Nagao, *Tetrahedron Lett.*, 1987, **28**, 6289; P. Magnus and J. Lacour, *J. Am. Chem. Soc.*, 1992, **114**, 767; P. Magnus, C. Hulme, and W. Weber, *J. Am. Chem. Soc.*, 1994, **116**, 4501; Y. Kita, H. Tohma, K. Hatanaka, T. Takada, S. Fujita, S. Mitoh, H. Sakurai, and S. Oka, *J. Am. Chem. Soc.*, 1994, **116**, 3684; Y. Kita, H. Tohma, T. Takada, S. Mitoh, S. Fujita, and M. Gyoten, *Synlett*, 1994, 427.
15. T. Kitawaki, Y. Hayashi, A. Ueno, and N. Chida, *Tetrahedron*, 2006, **62**, 6792.
16. K. Matsumoto, A. Tanaka, I. Yukio, N. Hayashi, M. Toda, and R. A. Bulman, *Heterocycl. Commun.*, 2011, **9**, 9.